Characterisation and Manipulation of Transition Metal Dichalcogenides *via* Atomic Force Microscopy



a thesis presented by Katie O'Neill

under the supervision of Prof. Georg S. Duesberg & Prof. Jonathan N. Coleman

> for the degree of Doctor of Philosophy in the subject of Physics

Trinity College Dublin The University of Dublin

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Declaration

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Characterisation and Manipulation of Transition Metal Dichalcogenides *via* Atomic Force Microscopy

Abstract

Scanning probe microscopy (SPM) has revolutionised nanotechnology and allowed the study and manipulation of materials at the nanoscale, making it ideal for the study of solid-state physics and semiconductor technologies. Atomic force microscopy (AFM) and scanning probe lithography (SPL) can be used as a 'toolbox' for mechanical treatments of various surfaces including polymers, metals and semiconductors. In recent years, two-dimensional (2D) layered materials, such as graphene and transition metal dichalcogenides (TMDs), have been heavily studied due to their high potential for use in a wide range of future nanoelectronic devices. Some semiconducting TMDs, such as MoS₂, are known to change their bandgap with decreasing layer thickness. Other TMDs, such as PtSe₂, have been shown to develop a band gap, i.e. go from semi-metallic to semiconducting.

The first results chapter of this thesis details where one such 'tool', the technique of nanoshaving, where materials are selectively removed by an AFM tip, is employed to produce nanopatterns of self-assembled monolayers (SAMs) on 2D materials. The materials used are monolayers of TMDs, namely MoS_2 and WS_2 non-covalently functionalised with a perylene derivative, perylene diimide (PDI). The approach involves rastering an AFM probe across the surface at a controlled increased load in ambient conditions. Due to the strong bond between PDI SAM and TMD, it is found that loads in excess of $1 \mu N$ are required to pattern the monolayer. Various pre-defined patterns including a grating pattern with feature sizes below 250 nm are demonstrated, showing the high precision of nanoshaving as an accurate and non-destructive lithographic technique for 2D materials. In addition, co-functionalisation of MoS₂ non-covalent is demonstrated using perylene-3,4,9,10-tetracarboxylic dianhydride (PTCDA) SAMs on previously nanoshaved areas. Work functions of the shaved heterostructures are also examined using Kelvin probe force microscopy, another SPM-based tool.

The second results chapter describes the use of SPL and SPM-based tools with the aim of producing a seamless, self-contacted device by way of suitable material, namely $PtSe_2$. A technique for manipulating TMD layers is explored and developed via nanomachining using AFM, which is used to reduce and control the layer thickness of $PtSe_2$ in the form of thermally assisted converted (TAC) films and mechanically exfoliated (ME) flakes. AFM and SEM results reveal the nanomachining of TAC films to be quite difficult due to the roughness of the films post-selenisation when compared to exfoliated materials. Nevertheless nanomachining TAC films results in thinner, smoother films overall. Machining of pristine Pt channels allows more control over the quantity of nanomachining, and also quality. For comparison with TAC films, electrically contacted ME PtSe₂ flakes are nanomachined at loads of $0.8 - 2.7 \,\mu$ N. Raman spectroscopy of the nanomachined area shows more defective/damage material in the remaining layers, which are more easily removed than pristine layers with successive passes of nanomachining at the same loads. This suggests a path for continuous monitoring of device performance with each layer removal down to the monolayer, starting with metallic multi-layers and thinning down to a semiconducting monolayer. This would enable the design of 'self-contacted' devices based on TMDs through the creation of a semiconducting channel via nanomachining with high mobility, low contact resistance and low power.

The research undertaken showed how AFM-based mechanical manipulation techniques, namely nanoshaving and nanomachining, can be used to examine and explore TMDs and TMD-based heterostructures, along with complementary spectroscopy techniques. In addition, this work has opened pathways to future studies on the capability of AFM as a toolbox for characterisation of 2D materials as well as SPL and 2D materials/heterostructures.

List of Publications

- K. O'Neill, R. Greig, R. Tilmann, L. Peters, C. P. Cullen, G. Cunningham, C. Bartlam, C. Ó Coileáin, N. McEvoy, & G. S. Duesberg. Patterning functionalised surfaces of 2D materials by nanoshaving.* *Nanomanufacturing and Metrology*, accepted. (2021)
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- D. Tyndall, S. Jaskaniec, B. Shortall, A. Roy, L. Gannon, K O'Neill, M. P. Browne, J. Coelho, C. McGuinness, G. S. Duesberg, & V. Nicolosi. Post-Synthetic Treatment of Nickel-Iron Layered Double Hydroxides for the Optimum Catalysis of the Oxygen Evolution Reaction. *npj 2D Materials & Applications*, 5, 73 (2021).
- S. Bhattacharjee, E. Caruso, N. McEvoy, C. Ó Coileáin, K. O'Neill, L. Ansari, G. S. Duesberg, R. Nagle, K. Cherkaoui, F. Gity, & P. K. Hurley. Insights into Multilevel Resistive Switching in Monolayer MoS₂. ACS Applied Materials & Interfaces, 12, 5, 6022-6029 (2020).
- 4. P. V. Shvets, D. Caffrey, K. Fleischer, I. Shvets, K. O'Neill, G. S. Duesberg, A. N. Vinichenko, K. Yu. Maksimova & A. Yu. Goikhman. Suppression of the metal-insulator transition in magnetron sputtered Ti₂O₃ films. *Thin Solid Films*, 694, 137642 (2020).
- 5. E. Coleman, S. Monaghan, F. Gity, M. Schmidt, J. Connolly, J. Lin, L. Walsh, K. Cherkaoui, K. O'Neill, N. McEvoy, C. Ó Coileáin, D. Buckley, C. O'Dwyer, P. K. Hurley & G. S. Duesberg. Large Area Growth of MoS₂ By Chemical Vapour Deposition. *ECS Meeting Abstracts*, MA2018-02 708 (2018).
- 6. W. Kim, H. Kim, T. Hallam, N. McEvoy, R. Gatensby, H. C. Nerl, **K. O'Neill**, R. Siris, G.T. Kim & G. S. Duesberg. Field-Dependent Electrical and Thermal Transport in Polycrystalline WSe₂ *Adv. Mater. Interfaces*, **5**, 11, 1701161 (2018).

List of Presentations

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- 5. *Scratching the Way to Faster Computers* SFI Thesis in 3 competition, TCD, Ireland 29th September 2017

Poster

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Contents

Ι	Intr	ODUCT	ION	Ι		
	I.I	Introdu	ction	I		
	1.2	Thesis (Dverview	5		
	1.3	Stateme	ent of Collaboration	6		
2	Тне	ory & B	ACKGROUND	7		
	2. I	Atomic	Force Microscopy	7		
		2.1.1	Contact (Static) Mode	9		
		2.1.2	Tapping/Non-contact (Dynamic) Mode	9		
		2.1.3	Lennard-Jones Potential	10		
		2.1.4	Spring Constant Calibration	13		
		2.1.5	Contact Mode Force Calibration (Lever Sensitivity)	14		
		2.1.6	Conductive AFM (C-AFM/PF-TUNA)	17		
		2.1.7	Kelvin Probe Force Microscopy (KPFM)	20		
		2.1.8	Nanomachining & Nanoshaving	24		
	2.2	Introdu	lection to 2D Materials	27		
	2.3	2D Mat	rerial Fabrication and Synthesis	32		
	2.4	Self-Ass	embled Monolayers (SAMs)	34		
	2.5	5 Layer-by-layer thinning methods				
	2.6	Device l	Fabrication	40		
		2.6.1	Schottky vs. Ohmic Contacts	42		
		2.6.2	Heterojunctions	44		
		2.6.3	Contact Resistance & Interface Geometries	45		
	2.7	Raman	Spectroscopy	47		
		2.7.1	Classical Raman Theory	48		
		2.7.2	Raman of TMDs	50		
	2.8	Photolu	minescence	52		
3	Experimental Methods 55					
	3.1	General	Materials	56		
	3.2	Materia	l Synthesis	56		
		3.2.1	Vapour Phase Deposition	56		
		3.2.2	Metal Deposition	57		

		3.2.3	Thermally Assisted Conversion (TAC)
		3.2.4	Chemical Vapour Deposition (CVD)
		3.2.5	Mechanical Exfoliation
	3.3	Lithogı	raphic Techniques
		3.3.I	Photo & Electron Beam Lithography
	3.4	Micros	copic Techniques
		3.4.I	Atomic Force Microscopy (AFM) 65
		3.4.2	Nanoshaving/Nanomachining
		3.4.3	Kelvin Probe Force Microscopy (KPFM)
		3.4.4	Conductive AFM
		3.4.5	Scanning Electron Microscopy
	3.5	Spectro	scopic Techniques
		3.5.1	Raman Spectroscopy
		3.5.2	Photoluminescence
	3.6	Electric	al Characterisation
4	Pat	ΓERNIN	g 2D Surfaces by Nanoshaving 73
•	4.I	Introdu	action
	4.2	Experir	nental Methods
	•	4.2.I	CVD of TMDs
		4.2.2	Surface Functionalisation of TMDs by SAMs
		4.2.3	AFM Characterisation & Manipulation
	4.3	Results	& Discussion
		4.3.I	Nanoshaving Pristine MoS ₂
		4.3.2	Nanoshaving PDI on MoS_2
		4.3.3	Further PL Analysis of Nanoshaved $MoS_2 & PDI \text{ on } MoS_2 & \dots & 85$
		4.3.4	Patterning PDI on MoS ₂ via Nanoshaving
		4.3.5	Co-functionalisation of MoS_2
		4.3.6	Change in Work Functions with Nanoshaving & Functionalisation 92
		4.3.7	Nanoshaving PDI on WS_2
	4.4	Conclu	isions
5	Nan	юмасн	uning $PtSe_2$ 97
,	٢. ١	Introdu	1ction
	5.2	Experir	nental Methods
	,	۲ 5.2.1	Thermally Assisted Conversion (TAC) Devices
		5.2.2	Mechanically Exfoliated (ME) Devices
		5.2.3	Characterisation & Manipulation
	5.3	Results	and Analysis
	,,	5.3.I	Nanomachining TAC PtSe ₂
		5.3.2	Nanomachining Pt Pre-Selenisation
		1.1	0

		5.3.3 Nanomachining Mechanically Exfoliated PtSe ₂ Flake Devices	I I 2
	5.4	Conclusion	121
6	Con	ICLUSIONS & FUTURE WORK	123
Ар	PEND	DIX A NANOSHAVING RAMAN SPECTRA	127
Re	FEREI	NCES	152

List of Acronyms

2D	Two-dimensional
AC	Alternating current
AFM	Atomic force microscopy
AM-KPFM	Amplitude modulated Kelvin probe force microscopy
BL	Bilayer
BP	Black phosphorus
C-AFM	Conductive AFM
СВ	Conduction band
CBM	Conduction band minimum
CMOS	Complementary metal-oxide-semiconductor
CPD	Contact potential difference
CVD	Chemical vapour deposition
DC	Direct current
DFT	Density functional theory
DOS	Density of states
EBL	Electron beam lithography
FET	Field effect transistor
FM-KPFM	Frequency modulated Kelvin probe force microscopy
номо	Highest occupied molecular orbital
HOPG	Highly oriented pyrolytic graphite
IPA	Isopropanol
InvOLS	Inverse optical lever sensitivity
KPFM	Kelvin probe force microscopy
LED	Light emitting diode
LJ	Lennard-Jones
LPE	Liquid phase exfoliation
LUMO	Lowest unoccupied molecular orbital
MBE	Molecular beam epitaxy
ME	Mechanical exfoliation/mechanically exfoliated
MESFET	Metal-semiconductor field effect transistor
ML	Monolayer
MOSFET	Metal-oxide field effect transistor
NTMD	Noble transition metal dichalcogenide
PF-TUNA	PeakForce tunneling AFM
PL	Photoluminescence
PSPD	Position-sensitive photodetector

RMS Root mean square	
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- SC Semiconductor
- SEM Scanning electron microscopy
- **SPM** Scanning probe microscopy
- **STM** Scanning tunneling microscopy
- SiO_2/Si 300 nm silicon oxide on silicon
 - TAC Thermally assisted conversion
 - TMD Transition metal dichalcogenide
 - **UHV** Ultra-high vacuum
 - UVL Ultraviolet photolithography
 - **VB** Valence band
 - **VBM** Valence band maximum
 - vdW van der Waals

Listing of figures

1.1	Summary of Microscopy techniques	3
2.1	Schematic of AFM in tapping mode	8
2.2	Illustration of PeakForce tapping cycle	I
2.3	Illustration of Lennard-Jones potential	2
2.4	Illustrations of force curves and cantilever motion	5
2.5	Schematics of PF-TUNA	8
2.6	Schematic of parallel plate capacitor setup used by Lord Kelvin	С
2.7	Schematics of nanomachining & nanoshaving	5
2.8	Periodic Table with TMDs highlighted	8
2.9	3D Schematics of TMD structures	9
2.10	Electronic band structure of MoS ₂	С
2.II	Electronic band structure of PtSe ₂	I
2.12	Schematic of 2D material fabrication methods	3
2.13	Functionalisation of 2D TMDs by SAMs	5
2.14	C-AFM patterning of BP	7
2.15	Selective chemical etching of MoS_2	7
2.16	Thinning of MoTe ₂ via laser irradiation	9
2.17	Diagram of MOSFET	I
2.18	Schottky junctions for p and n-type semiconductors	2
2.19	Heterojunction types	4
2.20	Schematic of contact geometry on 2D materials	6
2.21	Schematic of types of Raman scattering 49	9
2.22	Raman active vibrational modes of TMDs	I
2.23	Direct-gap exciton transitions of MoS_2	3
3.1	Photos of S & Se VPD furnaces 55	7
3.2	Schematic of two-zone furnace for TMD synthesis	С
3.3	Schematic of CVD synthesis & microreactor	I
3.4	Photo of custom-built ME transfer apparatus	2
3.5	Photos of maskless UV lithography IMP	4

	68 69 71 75 78
nctions	69 71 75 78
· · · · · · · · · · ·	71 75 78
	75 78
 	78
	'
	82
	83
	86
	88
	89
	91
	93
	94
	98
	101
	103
	106
	107
	108
	IIO
	III
	III
	I I 4
VL	115
passes)	116
)	117
	118
	I 20
)	121
)	121 128
)	121 128 128
)	121 128 128 129
	VL

For a research worker, the unforgotten moments of their life are those rare ones which come after years of plodding work, when the veil over nature's secret seems suddenly to lift, and when what was dark and chaotic appears in a clear and beautiful light and pattern.

Gerty Cori, biochemist & Nobel Prize winner

1 Introduction

1.1 INTRODUCTION

FROM THE EARLIEST TIMES, humans have been interested in the world of the small. From ancient Greek philosophers, such as Democritus - who proposed the concept of minute elements given the Greek word *atomos* meaning 'indivisible'¹ - to modern day particle physics, humans have been trying

to determine the building blocks of the universe. The question is, how do you visualise the world that is too small to see?

In 1967, anatomist and optical and electron microscopist, Savile Bradbury, noted the symbolic power of the microscope: 'Of all the instruments used by the scientist, the microscope is perhaps the one which most aptly symbolises this profession to the non-scientist.'² First coined in 1624 in Italy by the original Accademia Dei Lincei (which included Galileo),³ the term 'microscope' itself is derived from the Latin *microscopium* - literally 'an instrument for viewing what is small.'⁴ One of the earliest important works on microscopy was physicist Robert Hooke's (of Hooke's Law) *Micrographia* in 1665, containing large-scale finely detailed illustrations of specimens viewed under an optical microscope.⁵

For over 400 years, the optical microscope has primarily dominated the field. Nevertheless, microscopy has come a long way since the compound microscopes of the 17th century that Hooke used whereby two convex lenses produced a magnified image. As well as optical microscopy, other types of microscopy techniques based on electrons and atomic forces have come to the fore in the last century. Modern microscopy can be divided into three basic types: optical, electron (or ion) and scanning probe.

The optical microscope as we know it was developed in the mid-19th century. It uses visible light (wavelengths of $\sim 380 - 700$ nm) and transparent lenses to view objects on the micrometer scale (10^{-6} m) , making it ideal for red blood cells, human hair, etc. However, the fundamental limitations of light microscopy led to the development of the electron microscope in the 1931. Instead of light, electron microscopes use a beam of negatively charged particles (electrons) and electromagnetic or electrostatic lenses to view particles on the nanometer scale (10^{-9} m) , e.g. atoms. It was not until some 50 years later that scanning probe microscopy was developed and from that the atomic force microscope (AFM) was born. AFM produces images by 'feeling' the surface with a very sharp needle-like probe, instead of using light, electrons or lenses. SPM has revolutionised nanotechnology

and allowed the study and manipulation of materials on the nanoscale, making it ideal for the study of solid-state physics and semiconductor technologies.



Figure 1.1: Summary of common pros and cons of the three main techniques of microscopy.

The use of semiconductor (SC) technology has exploded after the development of the transistor,⁶ followed by the integrated circuit (IC) and the microchip. As a result, SCs have become ubiquitous in our everyday lives due to their highly economical miniaturisation and reliability, from transistors and diodes to microwave generators, solar cells and lasers, to name a few.

In 1965, Gordon Moore stated that the number of components per chip would approximately double every 2 years.⁷ Since the 1970's, the semiconductor industry has been able to maintain Moore's law through the implementation of increasingly complex manufacturing of silicon (Si) based transistors. This is despite its end being prophesied numerous times through quantum effects resulting in leakage currents at the nanoscale. Conventional silicon-based transistors with a channel width as small as 3 nm have been demonstrated, ⁸ with a report on the possibility of even single atom transistors, representing the ultimate physical limit of Moore's law.⁹ However, large scale integration

of these devices presents numerous challenges. Given current architectures, the devices would require major heat dissipation caused by contact resistance and leakage currents.

One way of potentially continuing Moore's Law is to utilise two dimensional (2D) materials, where there has been increasing interest in recent years thanks to modern advances in exfoliation techniques and particularly vapour-phase growth. These layered materials encompass semimetals, insulators, superconductors and semiconductors (SCs) such as graphene, hexagonal boron nitride (h-BN), transition metal dichalcogenides (TMDs), black phosphorus (BP), respectively, and many more. These materials all share a layered structure in common, whereby they can exist from a single layer of sub-nanometer thickness with significantly larger lateral sheet dimensions, to a bulk crystal consisting of countless layers held together by weak, out-of-plane Van der Waals (vdW) forces. Since Geim and Novoselov first isolated graphene from graphite in 2004, ¹⁰ 2D materials have been consistently in the research spotlight for applications in future nanoelectronics due to their fascinating optical¹¹⁻¹⁵, electrical¹⁶⁻¹⁸ and chemical/gas-sensing properties, ^{19,20} as well as ultrathin body and absence of dangling bonds.

As previously mentioned, SPM has revolutionised nanotechnology and allowed the study and manipulation of materials on the nanoscale, making it ideal for the study of 2D materials. However, for AFM in particular, little study has been carried on the use of its multitude of modes as a SPM-based 'toolbox' for the characterisation and manipulation of 2D materials. With AFM's ability to provide a whole host of tools *in-situ* to measure the mechanical and electrical properties as well as manipulation, 2D materials can be explored and characterised in a variety of ways that were previously not possible at the nanoscale.

1.2 THESIS OVERVIEW

The aim of this work is to take steps towards expanding the understanding of 2D materials and heterostructures through characterisation and manipulation using atomic force microscopy as the primary toolbox.

In **Chapter 1 Introduction**, the motivation behind this research is introduced and laid out, followed by an overview of this thesis. Thereafter is **Chapter 2 Theory & Background** involving the theoretical background of various AFM modes, i.e. the 'tools', including the manipulative techniques of nanoshaving and nanomachining, as well as the complementary Raman and photoluminescence spectroscopies. In addition, an overview of 2D materials, synthesis and device structures is discussed alongside a literature review of thinning processes.

Following the background and theory is **Chapter 3** pertaining to the experimental details for the synthesis and fabrication methods of the TMDs used in this work. In addition, characterisation techniques are introduced and discussed here.

The experimental results and discussion of this thesis are divided into two chapters. **Chapter 4 Patterning 2D Surfaces by Nanoshaving** explores the novel SPM technique of nanoshaving in patterning self-assembled monolayers (SAMs) on 2D group 6 TMDs. These organic-inorganic heterostructures are characterised further using Kelvin probe force microscopy (KPFM) and spectroscopic techniques such as Raman and photoluminescence (PL).

In **Chapter 5 Nanomachining PtSe**₂^{*} the less studied group 10 TMD, $PtSe_2$, is synthesised and fabricated into 2D devices. These layers are then removed mechanically by way of nanomachining, another SPM technique, with the aim of creating a semiconducting recess channel. These devices are characterised further with the complementary microscopic methods of conductive AFM, KPFM,

^{*}A small note on the structure of this thesis: to preserve the flow of the narrative, the results chapters, i.e. 4 and 5, are ordered chronologically in reverse.

SEM and Raman spectroscopy.

This thesis concludes with **Chapter 6** by reviewing the main findings, while subsequently suggesting directions for future investigations.

1.3 STATEMENT OF COLLABORATION

Some of the CVD MoS₂ samples shown in Chapter 4 were grown by Dr. Conor Cullen and Dr. Lisanne Peters in Trinity College Dublin. CVD WS₂ samples were grown by Dr. Graeme Cunningham in Trinity College Dublin. Some mechanically exfoliated samples shown in Chapter 5 were prepared by Max Prechtl and Rahul Dangi in the Universität des Bundeswehr in Neubiberg, Germany.

Young people, especially young women, often ask me for advice. Here it is, valeat quantum. Do not undertake a scientific career in quest of fame or money. There are easier and better ways to reach them. Undertake it only if nothing else will satisfy you; for nothing else is probably what you will receive. Your reward will be the widening of the horizon as you climb. And if you achieve that reward you will ask no other.

Cecilia Payne-Gaposchkin, astronomer & astrophysicist

2

Theory & Background

2.1 Atomic Force Microscopy

The atomic force microscope (AFM) was developed in 1986 by Quate, Gerber and Binnig, the latter of whom was one half of the team behind the Nobel winning scanning tunneling microscope (STM), which revolutionised surface science. AFM is a very high resolution scanning technique and one of the foremost tools for imaging, measuring, and manipulating matter at the nanoscale. It is primarily used to acquire spatially resolved data on surface features, such as the topography of the surface atomic layer, and also to measure forces at the surface. This information is gathered by exploring the specimen surface with a sharp tip (probe) on the end of a cantilever. Piezoelectric elements facilitate tiny but accurate and precise movements on electronic command and this enables high resolution scanning. Electric potentials of samples can also be measured using conducting cantilevers.²¹



Figure 2.1: Schematic of typical atomic force microscope (AFM) in tapping mode. The white arrow indicates rastering direction.

The following sections contain content adapted from the Bruker MultiMode 8-HR User Guide.²² The key operating principle behind the AFM is detection of the bending of the cantilever spring in response to external forces. This is clearly shown in Figure 2.1. When the tip is brought within close proximity of a sample surface, forces between the tip and the sample lead to a deflection of the cantilever. This deflection follows Hooke's Law (for small deflections) where the force applied to bend the cantilever is proportional to the cantilever displacement. By fixing the force/setpoint, the topographical data is measured by multiplying the measured displacement by the cantilever's spring constant. Typically, this deflection is measured using a laser spot reflected from the top surface of the cantilever into an array of photodiodes (see Figure 2.1).

Cantilevers are typically made of silicon or silicon nitride. The tip radius of curvature is on the order of nanometers and can be coated in a variety of materials depending on function, including diamond. In addition, tips can be terminated with other molecules to increase sharpness and sensitivity such as carbon nanotubes²³ and carbon monoxide.²⁴ The two primary modes of operation of an AFM are: contact and tapping/non-contact.

2.1.1 CONTACT (STATIC) MODE

This is the simplest mode and consists of the cantilever tip being dragged across the surface of the sample. The contours of the surface are measured directly using the deflection of the cantilever. It is essential that it be done in contact where the overall force is repulsive, as the strong attractive forces at the surface can cause the tip to crash into the sample. The feedback mechanism keeps the force between the tip and the sample surface constant during scanning by maintaining a constant deflection. This method, however, is prone to noise due to the static signal and can damage delicate surfaces, as well as the tip itself, when the tip is dragged across.

2.1.2 TAPPING/NON-CONTACT (DYNAMIC) MODE

An oscillating input signal is applied to make the cantilever (supporting a sharp tip at its end) vibrate up and down near its fundamental resonance frequency. This is done by a small piezoelectric transducer mounted in the AFM tip holder. The actual movement of the probe will depend on its interaction with the specimen surface when brought into close proximity. Changes in the vibrating frequency due to tip-sample interaction are used to maintain a constant distance from the surface while rastering across the sample. Images are produced by imaging the force of the intermittent contact of the tip with the sample surface while maintaining a constant force or force gradient by a feedback loop. The short-range interaction is usually described by the empirical Lennard-Jones type interaction potential. This method of 'tapping' lessens the lateral forces and consequently the damage done to the surface and the tip, compared with contact mode. This makes tapping mode useful for examining delicate samples. However in ambient conditions, most samples develop a liquid meniscus layer. Because of this, keeping the probe tip close enough to the sample for short-range forces to become detectable while preventing the tip from sticking to the surface presents a major problem to dynamic mode operations.²⁵

In this work, the primary mode used for topographical measurements was ScanAsyst-Air, a Bruker proprietary tapping mode. It utilises Bruker's trademarked PeakForce Tapping mechanism (Fig. 2.2) which, though similar to standard tapping mode, decouples cantilever response from resonance dynamics, to automatically adjust all critical imaging parameters. This means it operates in a non-resonant mode, with gain and setpoint constantly being re-adjusted automatically through a sophisticated real-time feedback loop allowing optimal data to be collected whilst minimising noise. Force data between the tip and the surface is acquired at each pixel.

2.1.3 LENNARD-JONES POTENTIAL

Proposed by Sir John Edward Lennard-Jones in 1924, the Lennard-Jones (LJ) potential describes the potential energy of interaction between two non-bonding atoms or molecules based on their distance of seperation.²⁶ It consists of two parts; a steep repulsive term representing Pauli repulsion, and a



Figure 2.2: Illustration of PeakForce tapping cycle in ScanAsyst-Air mode with probe motion and vertical force as a function of time. Blue indicates tip approach while red indicates retract. Below the graph are illustrations of the AFM probe as it approaches and interacts with a sample surface. Each image corresponds to the labelled portion of the force curve.²²

smoother attractive term, representing London dispersion forces, as illustrated in Fig. 2.3. The LJ potential frequently forms the base of many computational models and is given by:

$$V(r) = 4\varepsilon \left[\left(\frac{\sigma}{r}\right)^{12} - \left(\frac{\sigma}{r}\right)^6 \right]$$
(2.1)

which is sometimes expressed as:

$$V(r) = \frac{A}{r^{12}} - \frac{B}{r^6}$$
(2.2)

where V is the intermolecular potential between the two atoms or molecules, ε is the well depth and a measure of the strength of attraction between the two particles, σ is the distance at which the intermolecular potential between the two particles is zero, r is the distance of separation between both particles (as measured from their centers) and $A = 4\varepsilon\sigma^{12}$, $B = 4\varepsilon\sigma^{6}$. In addition, σ gives the van der Waals radius, which is equal to half of the internuclear distance between the two particles. In the case of AFM, the two interacting particles represent the tip-sample interactions.²⁷



Figure 2.3: Illustration of Lennard-Jones potential which is used to describe the tip-sample interaction in AFM.

2.1.4 Spring Constant Calibration

The spring constant of the majority of AFM probes is calibrated *via* the thermal tune method. This was first proposed by Hutter and Bechhoefer who noted that the spring constant of a cantilever could be related to its thermal energy when the law of equipartition is applied to the fundamental mode of the cantilever, resulting in: ^{28–30}

$$\frac{1}{2}k_0\langle z_c^2\rangle = \frac{1}{2}k_BT \tag{2.3}$$

where k is the spring constant of the cantilever, $\langle z_c^2 \rangle$ is the mean square deflection of the cantilever's thermal motion, k_B is the Boltzmann constant and T is the absolute temperature. This equation assumes that the cantilever acts as a perfect simple harmonic oscillator. To compensate for the fact that cantilevers do not act like these, Butt and Jaschke added a small correction to Eqn. 2.3:³¹

$$k = \frac{0.971k_BT}{\langle z_c^2 \rangle} \tag{2.4}$$

Butt and Jaschke also recognised that the optical lever deflection detection scheme detects cantilever inclination rather than true displacement, resulting in an additional error.³¹ This correction combined with an additional correction^{*} necessary if the cantilever is mounted at an angle α off horizontal³³ (as is wont to happen in AFM) results in the final formula:³⁴

$$k = \frac{0.817k_B T \cos^2 \alpha}{\langle z *_c^2 \rangle} \tag{2.5}$$

where the asterisk in $\langle z *_c^2 \rangle$ designates the 'virtual' deflection measured by the optical lever. The thermal vibrations of the mounted tip (once the laser is aligned and tip is lowered near the sample surface) are recorded by the photodetector, which has a finite bandwidth up to 2 MHz. The

^{*}These corrections assume a rectangular cantilever. Different corrections are needed when applied to v-shaped cantilevers, e.g. ScanAsyst-Air. ³²

original displacement time-series is Fourier transformed to segregate other broadband noise contributions from the narrowband thermal noise around resonance.²² Using Eqn. 2.5 to calculate the spring constant, the software automatically fits the resonance peak in the filtered data with a Lorentzian distribution for use in air:²²

$$A(\nu) = A_0 + \frac{C_1}{(\nu - \nu_0)^2 + C_2}$$
(2.6)

where $A(\nu)$ is the amplitude as a function of frequency ν , A_0 is the baseline amplitude, ν_0 is the centre frequency of the resonant peak, C_1 and C_2 are Lorentzian fit parameters. With the spring constant calibrated, typical tapping mode operation can be carried out using amplitude modulation detection with a lock-in amplifier.

2.1.5 CONTACT MODE FORCE CALIBRATION (LEVER SENSITIVITY)

To calculate the contact force between the tip and sample during Contact AFM, it is imperative to calibrate the contact force on a hard sample surface, i.e. negligible surface deformation should occur with the cantilever deflecting while the probe pushes against the surface. The sample surface must also be clean as contamination can result in force dissipation instead of the cantilever bending. Most commercial AFMs use an optical method to detect the deflection of the cantilever, i.e. topography is measured from the photodetector voltage and the inverse optical lever sensitivity (InvOLS), whereas the force is the product of the displacement in nm and the spring constant.

As previously described, a laser spot reflects from the back side of the cantilever onto a position-sensitive photodetector (PSPD), which yields a signal corresponding to tip/surface interactions. The deflection sensitivity allows conversion from the raw photodiode signal (in volts) to the deflection of the cantilever (in nm), which must be calibrated before accurate deflection data

can be obtained. This sensitivity is equal to the inverse of the force curve slope while the cantilever is in contact with the hard sample surface.

If the sensitivity is calibrated on a material much stiffer than the cantilever, the inverse value of the AFM's optical lever sensitivity (InvOLS, nm/V) is measured, i.e. how many volts of deflection signal are produced by a given deflection of the cantilever tip. To calibrate, the deflection error (nm) is plotted against height, Z (nm) to give the force-versus-distance curves (force curves for brevity), which are a graphical representation of the deflection error compared to Z piezo position. Deflection error is a comparison to the separation between the probe tip and the sample surface.

A force plot is an observation of tip-sample interaction that provides information regarding the sample and tip. In SPM, by combining force curves at regularly spaced intervals across a sample, a force map of the sample's electronic properties, elastic modulus and chemical bonding strengths can be obtained.³⁵



Figure 2.4: (a) The force curve represents the deflection signal for a complete extension-retraction cycle of the Z-piezo. **(b)** Illustrations of a cantilever probe as it approaches and interacts with a sample surface.²² Each numbered image corresponds to the numbered portion of the force curve.

Probes with lower spring constants are very sensitive to attractive and repulsive forces. In Fig. 2.4, the force curve represents the typical deflection signal for a complete extension-retraction cycle of the Z piezo. The numbers on the graph highlight each stage, which correspond to the numbered

illustrations describing the cantilever motion. The horizontal axis plots the tip movement relative to the sample with the vertical axis plotting the cantilever deflection.

- 1. The piezo extends and the probe descends towards the sample surface. There is no contact between the tip and the sample surface yet.
- 2. Attractive forces near the sample surface pull the tip down, i.e. snaps into contact.
- 3. As the tip presses into the surface, the cantilever deflects upward.
- 4. The piezo retracts and the probe ascends until the retraction forces are in equilibrium with the surface forces. The cantilever relaxes downwards until it is no longer deflected. The piezo continues to retract and the probe ascends further. The cantilever bends downward as the surface attraction holds onto the tip.
- 5. As the piezo continues retracting the probe continues to ascend. With no further contact with the sample surface, the tip is now free of the sample, thus the cantilever is neither deflected upward or bent downward.

Using the slope of the force curve, we get:

$$InvOLS = \Delta D = \frac{\Delta z}{\Delta V} \Delta V \tag{2.7}$$

The inverse of the optical lever sensitivity (InvOLS) comes from requiring the slope of the hard contact region to be equal to 1, as on a hard surface (eg. sapphire), only the cantilever deforms during indentation ($\Delta z = \Delta D$, where z is the position of the piezo in the axial dimension and D is the deflection of the cantilever).³⁶ The drawback of this method is that InvOLS is dependent on a wide variety of factors including laser spot size/shape,³⁷ position/optical geometry,³⁸ cantilever length, backside reflectivity,³⁹ possible cantilever twist/rotation due to lateral forces,³⁶ etc. Typical InvOLS

values range from 1 - 100 nm/V.

2.1.6 Conductive AFM (C-AFM/PF-TUNA)

In conductive modes of AFM, a stationary (DC) voltage bias is applied between the tip and sample and the corresponding DC electric current flow is measured, from which the conductance of the sample can be mapped locally. In addition, at specific locations on the sample, the dependence of conductivity on the applied bias can be measured by performing DC voltage sweeps. The applied voltage can be modulated slowly, enabling the differential conductance $\frac{dI}{dV}$ to be measured. With the Bruker Multimode 8, two modes of conductive AFM were used:

C-AFM

Conductive AFM (C-AFM) is suited to imaging samples spanning a wide range of conductivity. While scanning in contact mode, a linear amplifier with a range of 1 pA - 1 μ A senses the current passing through the sample. By maintaining a constant force between the tip and sample, topographic and current images are generated simultaneously, enabling the direct correlation of local topography with local electrical properties. However, the use of contact mode for topography has proven to be a major limiting factor, particularly for samples that require low imaging forces in either/both vertical or lateral directions, e.g. conductive polymers, loosely bound nanowires, etc.

PF-TUNA

Another conductive mode is PeakForce tunneling AFM (PF-TUNA), a Bruker proprietary mode combining PeakForce Tapping mode with a high-bandwidth, low-noise current amplifier. It is suitable for highly resistive samples as it can sense sub-pA tip/sample currents using a linear amplifier



Figure 2.5: (a) Schematic of PF-TUNA setup.²² **(b)** Diagram of a PF-TUNA tapping cycle with Z position, force and current as a function of time. Blue indicates tip approach while red indicates retract.²²

with a range of 80 fA - 120 pA. PF-TUNA also largely eliminates lateral forces that tend to damage the tip and/or sample during imaging in contact mode and allows high resolution nanomechanical information to be collected simultaneously with correlated nanoelectrical properties. A force curve is performed at every pixel and the corresponding graph of Force vs. Time is referred to as the 'heartbeat' (see Fig. 2.5(b)). From the force curves, mechanical properties of the sample such as adhesion, modulus and deformation can be determined.

Fig. 2.5(b) illustrates what happens when the periodically modulated PeakForce Tapping probe interacts with the surface. The top line represents the Z-position of the cantilever base, as a function of time, as it goes through one period. The middle line represents the force measured by the probe during the approach (blue) and withdraw (red) of the tip to the sample. The bottom line (green) represents the detected current passing through the sample. Since the modulation frequency is about 1 kHz, the time from point A to point E is about 1 ms. From the current-time plot, such as in Fig. 2.5(b), PF-TUNA extracts 3 measurements:

• Peak current is the instantaneous current at point C, coinciding with the peak force. It is
usually the maximum current measured.

- **TUNA current** is the average current over a full tapping cycle, from point **A** to point **E**. It includes both the current measured while the tip is in contact and off surface.
- **Contact current** is the average current only when the tip is in contact with the surface, from the snap-on at point **B** to the pull-off at point **D**.

2.1.7 Kelvin Probe Force Microscopy (KPFM)

Kelvin probe force microscopy (KPFM), a relatively new technique, was first developed by Nonnenmacher *et al.*⁴¹, with the name originating from the macroscopic method developed by Lord Kelvin in 1898 using a vibrating parallel plate capacitor setup (see Fig. 2.6).⁴⁰ While employing the same principle, KPFM is based on non-contact AFM (nc-AFM) and the electrostatic interaction is continuously minimised between the scanning tip and the surface. KPFM is a powerful tool used to investigate the local surface potential and work function of samples with high spatial



Figure 2.6: Schematic of vibrating parallel plate capacitor, consisting of copper and zinc plates, developed by Lord Kelvin in 1898 - the macroscopic method which modern Kelvin probe techniques are based on.⁴⁰

and energy resolution.⁴² While similar to electrostatic force microscopy (EFM) which also measures electrostatic properties, these force contributions are compensated in KPFM by applying a DC-bias voltage between the tip and sample. The contact potential difference (CPD) is determined for metals and SCs, which is related to the sample's work function (φ), while for insulators, information about local charges and dipoles is obtained.

There are two main operating modes in KPFM:⁴²

Amplitude modulation (AM-KPFM)

Amplitude modulation KPFM (AM-KPFM) is a two-pass procedure where the surface topography is collected in a tapping mode on the first pass and the surface potential is measured on the second, in

a lift mode.[†] The amplitude of the cantilever oscillation at the AC-frequency ω_{ac} is measured, which is proportional to and induced by the electrostatic force. If the tip and sample are at the same DC voltage, there is no force on the cantilever at ω_{ac} and the cantilever amplitude will go to zero. Local surface potential is determined by adjusting the DC voltage on the tip V_{tip} or sample V_{sample} until the oscillation amplitude goes to zero and V_{tip} or V_{sample} is equal to the surface potential. The voltage bias applied V_{DC} is recorded to construct a voltage map of the surface.

The limiting factor in this mode are large AC voltages can potentially induce band-bending at the surface of SCs, leading to an inaccurate determination of φ .⁴³

Frequency modulation (FM-KPFM)

Frequency modulation (FM-KPFM) is a tapping mode single-pass technique and does not use a lift mode, in contrast to AM-KPFM. It has a higher spatial resolution than AM-KPFM but a lower signal-to-noise ratio. An AC voltage with amplitude V_{AC} and frequency f_{AC} (angular frequency ω_{AC}) superimposed on a DC voltage V_{DC} is applied between the tip and sample. In the case of AFM, the most significant contribution is due to the forces perpendicular to the sample surface (z direction). The resulting electrostatic force is given by:

$$F_{el} = -\frac{1}{2} \frac{\partial C}{\partial z} (\Delta V)^2$$
(2.8)

where *C* is the capacitance and ΔV is the total voltage:

$$\Delta V = V_{DC} - \frac{\Delta \varphi}{e} + V_{AC} \sin(\omega_m t)$$
(2.9)

[†]A lift mode is defined as when the tip rasters at a set height above the surface, following the surface contours to maintain constant seperation.

and $\Delta \varphi$ is the CPD between the probe and sample.

Combining these wxpressions, Eqn. 2.8 can be separated into three terms:

$$F_{el} = -\frac{1}{2} \frac{\partial C}{\partial z} \left[\left(V_{DC} - \frac{\Delta \varphi}{e} \right)^2 \right] + \frac{\partial C}{\partial z} \left[V_{DC} - \frac{\Delta \varphi}{e} \right] V_{AC} \sin(\omega_m t) + \frac{1}{4} \frac{\partial C}{\partial z} V_{AC}^2 \cos(2\omega_m t)$$
(2.10)

The electric field gradient is given by:

$$\frac{\partial F_{el}}{\partial z} = -\frac{1}{2} \frac{\partial^2 C}{\partial z^2} \left[\left(V_{DC} - \frac{\Delta \varphi}{e} \right)^2 \right] + \frac{\partial^2 C}{\partial z^2} \left[V_{DC} - \frac{\Delta \varphi}{e} \right] V_{AC} \sin(\omega_m t) + \frac{1}{4} \frac{\partial^2 C}{\partial z^2} V_{AC}^2 \cos(2\omega_m t)$$
(2.11)

The applied AC voltage modulates the force and force gradient at frequencies ω_m and $2\omega_m$. With regards modulating frequency, Hooke's Law states:

$$F = k(z - z_0) \tag{2.12}$$

Taking the derivative, gives:

$$\frac{\partial F}{\partial z} = k \tag{2.13}$$

Thus the force gradient and spring constant are seen to be equivalent. The electrostatic force shifts the resonant frequency of a cantilever with effective mass m* as follows:

$$\omega_0' = \sqrt{\frac{k - \frac{\partial F}{\partial z}}{m*}} \tag{2.14}$$

The applied AC voltage modulates F_{el} and $\frac{\partial F_{el}}{\partial z}$ according to Eqn. 2.11.

In this work, using a Bruker Multimode 8, the mode used was PeakForce KPFM (PF-KPFM), a two-pass mode that is a combination of Bruker's proprietary PeakForce tapping mode and standard FM-KPFM. It measures surface potential/work function using a lift mode (like in AM-KPFM)

variation of FM-KPFM imaging while simultaneously measuring nanomechanical properties. This translates as the first pass in a standard PeakForce tapping mode trace and retrace with the tip in the second pass lifted to a specified height above the sample surface (non-contact) and measuring the surface potential. The two measurements are interleaved - that is, they are each acquired one line at a time sequentially with both images displaying simultaneously in real time.

Determination of Work Function (ϕ) with System Bias

In this work, the bias was applied through the sample as the CPD images of higher potential translate to higher work function, and vice versa. For a sample-biased system, the work functions, φ , can be determined from the CPD as such:⁴²

$$CPD = V_{probe} = \frac{\varphi_{sample} - \varphi_{probe}}{|e|}$$

$$\varphi_{probe} = \varphi_{Au} - |e|CPD_{Au}$$

$$\varphi_{sample} = \varphi_{probe} + |e|CPD_{sample}$$
(2.15)

Where e is the elementary charge. Equivalently, a bias can be applied to the tip, resulting in the following equations for work functions:⁴²

$$CPD = V_{probe} = \frac{\varphi_{probe} - \varphi_{sample}}{|e|}$$

$$\varphi_{probe} = \varphi_{Au} + |e|CPD_{Au}$$

$$\varphi_{sample} = \varphi_{probe} - |e|CPD_{sample}$$
(2.16)

Using a calibrated tip with a known φ , the work function of the sample can be calculated from the CPD using Eqn. 2.15. However, as the work function is highly sensitive to surface cleanliness,⁴¹ tip

shape and wear,^{44,45} sample quality and grounding and experimental conditions, operation under ultra-high vacuum (UHV) conditions is preferred for absolute work function values.⁴⁶

2.1.8 NANOMACHINING & NANOSHAVING

Scanning probe microscopy (SPM) techniques have been widely employed in the fabrication of nanostructures on a wide variety of surfaces.^{47–49}Though their throughput is limited, the nanoscale control and resolution afforded by scanning probe microscopy has prompted the development of a wide variety of scanning-probe-based patterning methods,⁴⁷ with some outperforming other lithography techniques. In addition, the ability of *in-situ* imaging gives AFM nanomachining and nanolithography an advantage over other lithographic techniques. Nanomachining uses the mechanical force exerted by the AFM tip to selectively remove material from a surface. It has been successfully applied to modify solid substrates, polymers and self-assembled monolayers.⁴⁷

Like nanomachining, nanoshaving is when an SPM tip is used to scratch a surface mechanically or electrochemically with a required normal force. ⁵⁰ It differs from nanomachining as its purpose is to selectively remove surfactants rather than penetrating the sample surface. ⁵¹ Mechanical displacement, or nanoshaving of self-assembled monolayers (SAMs) by an AFM probe is a relatively inexpensive technique, ideal for fundamental studies. ⁵⁰ The ability of *in-situ* imaging also gives AFM nanomachining/shaving an advantage over other lithographic techniques.

Whether nanoshaving or nanomachining, the contact force between the tip and the sample can be calculated once the spring constant and the delclection sensitivity is known. Using the cantilever spring constant, k, and Hooke's Law, the contact force is defined as:

$$F_{tip} = kd \tag{2.17}$$



Figure 2.7: (a) Schematic of AFM setup for nanomachining/shaving. **(b)** SEM image of wide-angle diamond cone tip used in nanomachining & nanoshaving (courtesy of Adama Innovations Ltd.). **(c)** Schematic of typical nanoshaving process. The adsorbed SAM is selectively removed from the aluminium oxide surface. ⁵⁰

where d = deflection sensitivity $(nm/V) \times deflection setpoint (V)$.

Despite being highly researched materials for a wide range of applications, to date, there has been little exploration of the nanopatterning of SAMs on 2D materials using AFM. These versatile material removal applications of AFM make it an excellent candidate for investigating noncovalently functionalised 2D systems and offer advantages over established techniques such as bottom-up approaches and focused ion beam (FIB) milling. ⁵²

The stability of the tip can be the limiting factor in creating reproducible patterns as tips are prone to deformation and contamination from debris of removed materials. A solution to the problem for this study was to use wear-resistant diamond tips with broad cone angles (see Fig. 2.7(b), courtesy of Adama Innovations Ltd.), the contact size can be well characterised and stays constant during repeated patterning.

The mechanical nature of the nanomachining/shaving process is the root of its most significant limitations. The parameters must be carefully tuned to ensure controllable and reducible material

removal or thinning. In the case of nanoshaving, the SAM adsorbates (resist) should be selectively removed without substantially damaging the underlying 2D material. Different factors such as sample material, tip speed, applied force, tip angle, sliding direction, humidity and sample orientation affect the mechanism by which the material is removed, as well as the quality and quantity of removal.⁵³ Furthermore, AFM-based mechanical nanomachining/shaving has the disadvantage of ridge formation or the accumulation, so-called 'pile-up', of materials around the features.⁵⁴

2.2 INTRODUCTION TO 2D MATERIALS

Graphene initially sparked huge interest for 2D materials to be used for logic electronic applications. However, graphene's lack of band gap has meant that it is very difficult to implement in logic devices, particularly field effect transistors (FETs), despite its other extraordinary properties.^{11,55} TMDs are a class of layered materials with the general formula of MX₂, where M is a transition metal (eg. Mo, W, Pt) and X is a chalcogen (eg. S, Se, Te). Considerable interest has developed in the most popular TMDs (group VI), where MoS_2 in particular is expected to substitute silicon for certain applications in the electronics industry. 56-59 Research into the dichalcogenides of Mo and W in particular has exploded, due to their excellent physical properties such as thickness-dependent semiconducting/metallic behaviours, 11 tuneable band gaps, $^{60} \sim 250 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$ carrier mobility at RT^{II} and outstanding on/off ratio in FETs (~ 10⁸).⁶¹ At present, there are over 40 TMDs known.⁶² The electrical properties of 2D TMDs have be shown to be dramatically modulated by thickness, structural phase transitions, strain-engineering and heterostacking. Also, by changing the chalcogen species, the electronic structure is affected more profoundly than the substitution of metal atoms. However a trend can still be observed whereby the d bands broaden with increasing atomic number of the chalcogen, resulting in a decrease in band gap. For example, the band gaps of $MoS_2 > MoSe_2 > MoTe_2$, from 1.3 to 1.0 eV respectively.⁶³

In TMDs, each metal atom is covalently bound to 6 chalcogen atoms (six-fold coordinated) and each chalcogen is covalently bound to 3 metal atoms (three-fold coordinated) in either a trigonal prismatic, octahedral or distorted octahedral configuration. These then form monolayer structures referred to as 2H, 1T and 1T' respectively, ⁶³ which are held together by weak van der Waals forces. ^{11,16} Monolayer (ML) MoS₂ forms a stable 2D trigonal prismatic 2H structure. ⁶⁴ This lattice configuration results in a D_{6b} point-group symmetry with three atoms in a unit cell, which can be see in Fig. 2.9. Group VI



Figure 2.8: The periodic table of elements with transition metals (TM) and chalcogens (C) for layered TMDs highlighted. Elements outlined in red are the TMDs studied in this work.

TMDs tend to be most stable in the 2H phase, whereas those with a group X TM centre are more stable in the 1T octahedral configuration.⁶⁵ The 2H-phase TMDs tend to be semiconducting while many of the TMDs in the 1T phase tend to be metallic or semimetallic in the bulk.

While bulk MoS_2 and WS_2 in the 2H phase are indirect bandgap SCs, the MLs of these materials show direct and wide bandgaps.¹⁶ In the bulk, MoS_2 and WS_2 have a bandgap of 1.2 and 1.3 eV, and these become direct bandgaps in the ML of 1.8 and 2.1 eV, respectively.^{66–69}

Fig. 2.10 shows the calculated electronic bandstructure of $2H-MoS_2$. The indirect bandgap in bulk MoS_2 (Fig. 2.10(a)) results in the direct excitonic transitions occurring at high energies at the K point. With reduced layer thickness, the indirect bandgap becomes larger, while the direct excitonic transition barely changes. For ML MoS_2 in Fig. 2.10(d), it becomes a direct bandgap semiconductor. This dramatic change of electronic structure in ML MoS_2 can explain the observed jump in ML photoluminescence efficiency.⁶⁶

Like MoS₂, ML WS₂ also forms a stable trigonal prismatic 2H structure, resulting in a D_{6b} point-group symmetry.⁷⁰ Similarly, in the ML, the conduction band minimum (CBM) and the



Figure 2.9: Three-dimensional schematic representation of the layered structures studied in this thesis of 2H-MoS₂, 2H-WS₂ and 1T-PtSe₂ (1T) along with their associated stacking configurations and unit cells. The crystallographic directions are indicated. The metal atoms, Mo, W and Pt, are shown in lilac, navy and silver respectively, with chalcogens S as yellow and Se as orange.

valence band maximum (VBM) are located at the K point. When the layer number increases, the VBM shifts towards the Γ point while the CBM moves to between the K and Γ point.⁷¹



Figure 2.10: Calculated electronic band structures of (a) bulk MoS_2 , (b) quadrilayer MoS_2 , (c) bilayer MoS_2 , and (d) monolayer MoS_2 .⁶⁶ The solid arrows indicate the lowest energy transitions.

A layered material less studied than group 6 TMDs is PtSe₂, which has been shown to develop a band gap (i.e. go from semi-metallic in the bulk phase to semiconducting in a single layer) with decreasing thickness of films.^{72,73} Unlike the more commonly examined Mo or W dichalcogenides, PtSe₂ differs in crystal (Fig. 2.9) and band structure (Fig. 2.11). ML PtSe₂ forms a dynamically stable 2D octahedral 1T structure, more commonly known as the CdI₂ crystal group, which also comprises of other TMDs such as HfS₂ and SnS₂.^{74,75} This CdI₂ lattice type configuration of PtSe₂ has a D_{3d} point-group symmetry with three atoms in a unit cell. The ML and bilayer (BL) have been reported as having indirect band gaps of 1.2 – 1.6 eV and ~ 0.8 eV, respectively.^{75–78} While ML and BL 1T-PtSe₂ are indirect SCs, bulk and few layer 1T-PtSe₂ are semi-metallic in nature, making it a very interesting candidate for the possibility of 'self-contacting' FETs. Zhao *et al.* reported that few-layer PtSe₂ FETs exhibit high electron mobility at room temperature (RT) (~ 210 cm²V⁻¹s⁻¹)

on SiO₂/Si substrate.⁷⁹ Nevertheless calculations predict significantly higher mobilities of ~ 4000 cm²V⁻¹s⁻¹ for PtSe₂ at RT.⁷⁸



Figure 2.11: Calculated electronic band structure curves of (a) monolayer PtSe₂, (b) bilayer PtSe₂ and (c) bulk PtSe₂.⁷⁶ The arrows indicate the lowest energy transitions.

ML PtSe₂ is relatively robust in nature as its structure changes negligibly under biaxial tensile deformation,⁷⁵ making it a suitable candidate for nanomachining. However, it exhibits an extraordinarily strong interlayer interaction, which is significantly stronger than other TMDs (e.g. MoSe₂, WSe₂), presenting a challenge in its use.⁸⁰ This is due to it being rich with *d* electrons (eg. Group 10), which are prone to d^2sp^3 hybridisation, forming the 1T phase.^{62,81} Other TMDs, like MoS₂ and WS2, tend to form d^4sp hybridisation, resulting in the 2H-phase. The intralayer hybridization between the *d* band of the transition metal and the p_z band of the chalcogen atom in TMDs is weak in d^2sp^3 but strong in the d^4sp configuration. The relatively weak hybridisation of d^2sp^3 results in stronger coupling between the two p_z bands of two intermolecular Se atoms, in the case of PtSe₂. This then leads to the extraordinarily strong interlayer interactions with nearly isotropic in-plane and out-of-plane mechanical interlayer coupling.⁸¹

2.3 2D MATERIAL FABRICATION AND SYNTHESIS

There are several commonly used methods for fabricating and synthesising 2D materials, each with associated advantages and disadvantages depending on the intended application, which are summarised in Fig. 2.12. These techniques can be generally categorised into three main classifications: mechanical exfoliation (ME), liquid phase exfoliation (LPE) and chemical vapour deposition (CVD).

Most pristine ML TMDs were initially achieved *via* mechanical exfoliation (ME), inspired by Novoselov and Geim's success with graphene.^{10,82} A relatively simple and straight-forward technique, ME uses adhesive forces through Scotch tape to peel highly crystalline weakly-bonded vdW layers apart repeatedly.^{83,84} Through this method, high-quality, pristine and macroscopic flakes of 2D materials can be produced, which are ideal for fundamental studies exploring properties. However, the disadvantages to this method are its low, unreliable throughput and tedious nature, particularly if the layered material has quite strong interlayer attraction. ME is completely inapplicable for production on an industrial scale. Nonetheless, it is invaluable in allowing new ideas and concepts for 2D materials to be explored.

On the opposite end of the scalability spectrum there is liquid phase exfoliation (LPE). As the name would suggest, this exfoliation method involves dispersing the bulk material in a liquid, primarily through sonication or shear mixing with an appropriate solvent.^{86–89} Through either method, large amounts of few-layer sheets of varying size and thickness can be produced. These few-layer flake dispersions are very suitable for production on an industrial scale, particularly for printable inks⁹⁰ and composites.⁹¹ Nevertheless, due to the large distribution in size and thickness, 2D materials made in this fashion are not suitable for high-quality electronic applications. In addition, problems can emerge due to re-aggregation of MLs upon drying and increased contact resistance between the flakes.



Figure 2.12: Schematic representation and summary of common pros and cons of the three main methods of 2D material fabrication.^{13,85} (CVD image courtesy of Dr. Christian Wirtz)

Both ME and LPE are *top-down* approaches for fabrication of TMDs. A *bottom-up* synthesis is chemical vapour deposition (CVD). This technique generally involves flowing a precursor gas/gases into a chamber to interact with one or more heated substrates.⁹² Rather than being reduced to few-layer or ML from bulk crystal, the TMD layer is synthesised from its base components. CVD-grown TMD films are generally of high-quality and can be deposited over large areas in ML (> 500 μ m²) or few-layers. Compared to ME and LPE, CVD is most suited to the current electronic manufacturing and is used extensively in the production of electronic devices.^{93,94} However, the caveat is that CVD can involve high temperatures (~ 400 – 1000 °C) that may not be compatible with the thermal budget of SC fabrication processes, as well as often toxic by-products. Another variant of CVD is thermally assisted conversion (TAC) or thermally assisted chalcogenisation, where a gas phase precursor is reacted with a solid phase, pre-deposited metal or oxide film.^{74;94:95} In the case of TMDs this would be a reaction of its components, a transition metal film with a gaseous chalcogen, to form the corresponding TMD, eg. a Pt film and Se forming PtSe₂. TAC is a simple process with the thickness of the resulting material easily controlled by the deposited

thickness of the metal film. Metal films can be deposited in several ways including sputtering and evaporation. The conversion process results in the production of large-scale polycrystalline TMD with grains of few- to multilayer thicknesses on the order of 10 - 100 nm.⁹⁴ Device geometry can be determined pre-conversion as TAC is highly compatible with lithographic techniques, such as photolithography and electron beam lithography for patterning. In addition, like CVD, TAC can be readily translated to current electronic manufacturing, though its applications are limited due to the high polycrystallinity in the converted films.^{72,96}

2.4 Self-Assembled Monolayers (SAMs)

Chemical functionalisation of layered 2D materials, such as graphene or transition metal dichalcogenides (TMDs), is essential in modifying their properties and expanding their capabilities and applications.⁹⁷ Exploiting the functionalisation of TMDs is seen as a route to their widespread application of TMDs in photonics, energy storage and conversion, drug delivery and medical devices.^{98,99} TMD surfaces tend to be rather inert to chemical functionalisation, due to saturation of basal plane chalcogen atoms.^{98,100} Possible routes to functionalisation of TMDs include the physisorption of molecules from vapour phase or solution-cast molecular assemblies.^{101,102} Self-assembled monolayers (SAMs), i.e. dense 2D monolayers comprised of organic molecules packed together and chemically adsorbed onto a surface,¹⁰³ have been in the scientific spotlight in recent years in various areas of research such as nano-functionalisation, biosensors and molecular electronics as active materials or insulators,^{101,104–106} and are a well-known technique for creating highly functional nanostructures on 2D surfaces.



Figure 2.13: Functionalisation of 2D TMDs by perylene-based SAMs: (a) AFM height profiles of the TMD/PBI heterostructures with coloured regions for the PBI multilayer (blue), PBI monolayer (red) and MoS₂ monolayer before functionalistion. (b) Schematic of a PBI SAM deposited on a MoS₂ monolayer.¹⁰²

Lee et al. demonstrated the tuning of graphene's electrical properties through the growth of a stable SAM of alkylsilane on top.¹⁰⁶ Kawanago et al. also applied SAM-based gate dielectrics to fabricate MoS₂ field-effect transistors.¹⁰⁷ Most recently, Tilmann et al. showed the non-covalent functionalisation of layered 2D materials through the selective formation of perylene-based SAMs on TMDs directly on the growth substrate (see Fig. 2.13),¹⁰² opening up a pathway to controllable and versatile functionalisation of 2D materials.

2.5 LAYER-BY-LAYER THINNING METHODS

As the electronic properties of 2D materials are thickness dependent, the ability to control the layer thickness affords control over the properties. Since 2D materials have become a significant topic of interest, many thinning methods have been proposed, such as plasma etching, surface oxidation, scanning probe lithography/machining and laser irradiation.

Though the throughput is limited, the nanoscale control and resolution and the ability of *in-situ* imaging gives AFM nanolithography an advantage over other lithographic techniques. One of the earliest instances of nanomachining 2D materials was by Lieber and Kim¹⁰⁸, whereby MoO₃ (\geq 5 nm) grown on MoS₂ was selectively machined leaving the MoS₂ layer underneath relatively intact, as the MoS₂ layer does not wear under the same conditions as MoO₃. Using a variant of this method, Hong *et al.* mechanically exfoliated Bi₂Se₃ nanoribbons (> 50 ML thickness) down to a single ML by drawing an AFM tip horizontally across the nanoribbons.^{109,110} Through the tip force applied parallel to the layers, the Bi₂Se₃ breaks along the vdW bonds, leaving some residual layers intact on the substrate.

Similar to nanomachining, oxidation-scanning probe lithography (o-SPL)⁴⁷ is another high-resolution technique. Rather than using mechanical force to etch the layers, it is based on the spatial confinement of an anodic oxidation reaction between the tip and the sample surface. For example, Liu *et al.* took advantage of the high spatial resolution of SPL and the tendency of black phosphorus (BP) to oxidise readily in ambient conditions to perform layer-by-layer thinning using conductive AFM (C-AFM), which was achieved via through DC local anodic oxidation (LAO).¹¹¹ By controlling the amplitude of the bias voltage during C-AFM (contact mode) scanning, patterns were created on a freshly exfoliated BP crystal, resulting in an oxdised byproduct easily removed by rinsing in water (see Fig. 2.14).



Figure 2.14: Tapping mode AFM images of patterned letter 'N' (a) after DC C-AFM patterning of BP and (b) after water rinsing.¹¹¹

In a process similar to LAO, Donarelli *et al.* used a conductive AFM tip, negatively biased, to scan a MoS_2 flake 10 nm above the surface in the desired pattern, with the silicon substrate grounded.¹¹² The process was carried out in ambient conditions, resulting in a water meniscus forming between tip and sample whereby the top MoS_2 layers oxidise to MoO_3 , which can be removed by way of selective chemical etching (See Fig. 2.15).



Figure 2.15: From left to right: MoS₂ pristine flake; MoS₂ flake after the lithography process (white dashed line represents the AFM tip path); MoS₂ flake after HCl etching. Tip voltage = -10 V; tip speed = $1 \mu m s^{-1}$. Etching parameters: 10 s in 0.05 M HCl.¹¹²

Another thinning method based on the oxidation of the 2D material surface is that of plasma oxidation. Li *et al.* used this technique on WSe₂ by exposing it to a remote oxygen plasma, resulting

in selective oxidation of the topmost layer, which can be easily removed in KOH solution. The oxidation is self-limiting due to the low kinetic energy of the oxygen radicals in the remote plasma, even with varying exposure times.¹¹³ This was again observed in MoS₂¹¹⁴ and MoTe₂¹¹⁵ (see Fig. 2.16) via the same method by Zhu *et al.* and Zheng *et al.* respectively, whereby the oxidised top layer MoO₃ was removed by thermal annealing instead of solution.

Similarly, 2D materials can also be thinned using laser irradiation. Laser thinning offers several advantages compared to other thinning methods, particularly simplicity, low cost and the lack of need for complicated lithographic processing.¹¹⁶ Castellanos-Gomez *et al.*¹¹⁷ reported thinning MoS₂ samples to MLs from bulk material with an initial thickness of ~ 20 layers, but without any control over layer precision during thinning. Hu *et al.*¹¹⁸ did show some thickness control in MoS₂, but were limited to producing thicknesses of three layers or below and needed multiple laser scans to achieve this control. Finally, Nagareddy *et al.* reported the humidity-controlled uniform thinning of multilayered MoTe₂ films down to the ML with layer-by-layer precision via continuous wave laser irradiation of an ultralow laser power density of 0.2 mW μ m⁻² (see Fig. 2.16).¹¹⁶



Figure 2.16: (a) Schematic illustration of the experimental setup used for thinning and in situ imaging of MoTe₂. (b) Change in thickness of multilayered MoTe₂ samples as a function of incident laser power at different humidity levels. (c) Optical microscope images and corresponding AFM topography images of MoTe₂ samples after each thinning cycle. (d) Schematic illustrations explaining the suggested MoTe₂ photochemical thinning mechanism.¹¹⁶

2.6 Device Fabrication

As previously mentioned, the use of semiconductor (SC) technology exploded after the development of the transistor,⁶ followed by the integrated circuit (IC) and the microchip. As a result, SCs have become ubiquitous in our everyday lives due to their highly economical miniaturisation and reliability. In recent years, 2D materials have become attractive for integration into current and future SC manufacturing architectures, such as transparent, flexible devices.

Of the many types of transistors available, field effect transistors (FETs), specifically metal-oxide-semiconductor FETs, have been the most important for digital electronics. A field effect transistor (FET) is a three-terminal device (source, drain, gate) in which the conductivity of a SC layer is modulated by a transverse electric field¹¹⁹ and is commonly used for weak-signal amplification. A FET consists of a channel of n- or p-doped SC material through which current can flow with the gate modulating the flow of charge carriers between the source and drain.¹¹⁹ They are known as unipolar transistors since they only involve single-carrier-type operation, wither electrons or holes. FETs can be subdivided into several types of devices such as junction (JFET), metal-oxide-semiconductor (MOSFET), metal-semiconductor (MESFET) and tunneling (TFET). Of these, the MOSFET is the most important device for advanced integrated circuits due to their low power consumption, massive manufacturing yield and can be constructed entirely out of silicon and its oxides.¹²⁰ As a result, the MOSFET is extensively used in the study of SC surfaces.¹²¹

The MOSFET consists of an MOS capacitor and two p-n junctions (see Fig. 2.17) and can be built as either n- or p-MOS transistors, depending on the polarities of the bulk (also known as the channel), source and drain regions. Fig. 2.17 shows an n-MOSFET whereby the majority carriers are electrons, the Si substrate is p-doped with n-type regions called the source and drain, and *vice versa* for a p-MOSFET. The gate oxide is a very thin insulating oxide layer that covers the channel region, on top of which a conducting gate material, either metal or highly doped polysilicon, is deposited.



Figure 2.17: A cross-section of a basic (n-)MOSFET structure.¹²²

For an n-MOSFET, when no voltage is applied between gate and source ($V_G = 0$), some current flows due to the voltage between the drain and source (V_{DS}). At relatively small values of V_{DS} , the IV characteristics of the device are linear, with the drain current I_D increasing proportionally with increasing drain voltage V_D . When a positive gate voltage is applied ($V_G > 0$), the minority carriers in the channel, in this case holes, are repelled and the majority carriers (electrons) are attracted to the gate oxide layer, increasing the electron density in the channel inversion layer. This allows a certain amount of I_D to flow through source to drain, allowing V_G to control the device current.

2.6.1 Schottky vs. Ohmic Contacts

of One the most common junctions within electronic devices are metal to SC interfaces, which often result in Schottky barriers, a potential energy barrier for electrons formed at a (rectifying) metal-SC junction. A metal-SC interface that has negligible contact resistance relative to the bulk is known as an ohmic contact (non-rectifying). It does not degrade device performance and can pass the required current with only a very small voltage drop when compared to the overall drop across the active region of the device.

A MESFET comprises of a non-rectifying metal-SC junction.



Figure 2.18: Energy band diagrams of metal n-type and p-type semiconductors under different biasing conditions: **(a)** thermal equilibrium; **(b)** forward bias; and **(c)** reverse bias. ¹²³

MESFETs have similar current-voltage characteristics to MOSFETs however the source and drain contacts are ohmic rather than p-n junctions like in MOSFETs. Current transport is also due to majority carriers unlike MOSFETs which are minority carrier devices. Schottky diodes are used in high frequency electronics due to their higher switching speeds.¹²³

If any two conducting materials in electrical contact are in thermal equilibrium, their Fermi levels must be continuous across the junction. These requirements determine a unique energy band diagram for the ideal metal-semiconductor contact (Fig. 2.18), where the barrier height $q\varphi_{Bn}$ is the difference between the metal work function φ_m and the SC electron affinity χ_s :

$$q\varphi_{Bn} = q\varphi_m - q\chi_s \tag{2.18}$$

Similarly, for an ideal contact between a metal and a p-type SC, the barrier height $q\varphi_{Bp}$ is:

$$q\varphi_{Bp} = E_g - (q\varphi_m - q\chi_s) \tag{2.19}$$

where E_g is the bandgap of the SC. Therefore, for any given SC and metal system, the sum of the barrier heights for n-type and p-type is expected to be equal to the band gap:

$$q(\varphi_{Bn} + \varphi_{Bp}) = E_g \tag{2.20}$$

On the SC side, the electrons in the conduction band (CB) encounter a built-in potential (V_{bi}) when trying to move into the metal:

$$V_{bi} = \varphi_{Bn} - V_n \tag{2.21}$$

where qV_n is the distance between the bottom of the CB and the Fermi level. Similar equations can be derived for a p-type SC.

For Schottky diodes, which utilise Schottky barriers, operated at RT, the dominant transport mechanism is thermionic emission of majority carriers from SC over potential barrier into the metal. When a forward bias V_F is applied to the contact, the electrostatic potential difference across the barrier is reduced, resulting in electron flow out of the SC and into the metal. Nevertheless, the flux of electrons from the metal into the SC remains the same because the barrier φ_{Bn} remains at its equilibrium value. Likewise when a reverse bias V_R is applied, φ_{Bn} remains constant and it is the flux of electrons from the SC into the metal that decreases below its equilibrium value.

While Schottky barriers are useful for certain applications, they can significantly hinder device performance of 2D materials.

2.6.2 Heterojunctions

A heterojunction or heterostructure is the interface between two regions or layers of materials that are juxtaposed. Heterojunctions are widely used as a base in SC devices owing to their structural, interfacial and electronic properties.¹²⁴ Heterostructures can be easily synthesised by availing of vdW coupling in 2D materials, e.g. sequentially stacking layered materials such as TMDs, resulting in a vertical heterostructure. They can also be fabricated through the adherence of self-assembled monolayer (SAMs) on top of 2D materials, creating organic-inorganic heterostructures.

Heterojunctions in SC materials result in unequal band gaps, that is the energy difference between the valence band (VB) and the conduction band (CB). The band gap can range from 0 eV (i.e. no gap) in a metal to over 4 eV in an insulator. Due to unequal band gaps, SC interfaces can be classified into three heterojunction types, which are illustrated in Fig. 2.19.



Figure 2.19: The three different types of heterojunctions in SC interfaces; type I - straddling gap, type II - staggered gap, and type III - broken gap.

Type I - straddling gap

The bandgap of one SC is entirely contained within the bandgap of the other one, i.e. $E_{g_1} > E_{g_2}$, $E_{V_1} < E_{V_2}$ and $E_{C_1} > E_{C_2}$. These junctions result in ultrafast recombination between electrons and holes and are mostly used in optoelectronic applications. Examples of 2D material heterostructures of this kind are MoS₂/ReS₂ and WSe₂/BP.^{125,126}

• Type II - staggered gap

The bandgaps overlap but the CB and VB of the second SC are both lower than the CB and VB of the first, i.e. $E_{V_1} > E_{V_2}$ and $E_{C_1} > E_{C_2}$. Type II heterojunctions can promote charge carrier transfer and separation at the interface and localised VBM and CBM states further enhance charge separation and enable carriers to be easily collected, making these junctions ideal for solar cells.¹²⁷ Staggered gaps can be found in heterostructures of the most commonly synthesised TMDs such as MoS₂, WS₂, MoSe₂ and WSe₂.¹²⁸

• Type III - broken gap

The bandgaps do not overlap at all or the VB of the first SC overlaps with the CB of the second SC. In the case of the latter, the CB electron states in the n-type material are more or less aligned with the VB hole states in the p-type material. This overlap at the junction results in a negative resistance, allowing for high electron tunnel efficiency. Broken gaps has been shown for Bi_2Te_3/WSe_2 and p-WSe_2/n-MoO_x vdW systems.¹²⁹

2.6.3 Contact Resistance & Interface Geometries

A major obstacle in the development and integration of practical 2D electronic and optoelectronic devices which must be addressed is that of reliable, low-resistance electrical contacts to 2D materials.¹³⁰ It has been shown that large contact resistances can occur between TMD channels and bulk metal contacts, significantly reducing the efficiency of current flow.¹³¹ Regardless of the

electrode material used, it requires the transmission of electrons through the interface between two different crystal lattices, which always causes a contact resistance due to reflections, Schottky barriers and scattering at the interface.¹³² Low contact resistance in 2D SC devices is crucial for achieving high 'on' current, large photoresponse¹³³ and high-frequency operation.¹³⁴ Nevertheless, the chemical interaction at the metal-SC interface governs everything for 2D materials. Pristine surfaces of 2D materials (i.e. no dangling bonds) hinder formation of any interface bonds with a metal, thus increasing contact resistance.¹³⁵

There are two possible geometries at the interface between bulk (3D) metals and 2D materials: top contact and edge contact (Fig. 2.20), each of which perform differently. While top contacts are relatively straightforward to fabricate, edge contacts are difficult to make using standard lithographic techniques on a single/few-layered 2D material.¹³⁵ However, in most experiments, the contacts are a combination of both geometries. This is particularly true for devices based on TAC-grown channels.



Figure 2.20: Schematic representation of interface geometries between 2D materials and their metal contacts in (a) top contact and (b) edge contact configurations.¹³⁵

As it is difficult for metals to covalently bond with pristine 2D material surfaces, a van der Waals (vdW) gap forms at the interface, acting as an additional tunneling barrier for the charge carriers, in addition to any innate Schottky barrier.^{130,136} This extra barrier reduces the charge injection from

metals, which leads to higher contact resistance. By reducing this vdW tunnel barrier as well as optimising edge contacts with all layers of the 2D material to produce a more seamless contact design, the contacts can be significantly improved with reduced resistance.

2.7 RAMAN SPECTROSCOPY

As a characterisation technique, Raman spectroscopy is powerful, non-invasive and non-destructive. It can provide crystal structure information, hence it is used extensively on 2D materials such as graphene and TMDs.⁷⁶ It works on the principle that when monochromatic light interacts with a sample, it will be absorbed, reflected or scattered. The vast majority of photons scatter elastically, known as Rayleigh scattering. These scattered photons are the same frequency, ω_0 , as the incoming beam. However, about one in 10⁷ are scattered inelastically in a process known as Raman scattering, due to the interactions with molecular or lattice vibrations in the sample, whereby the frequencies of the weakly scattered photons are $\omega_0 \pm \omega_q$, with ω_q is the vibrational frequency of the molecule. The Raman effect is named after the Indian physicist Sir C. V. Raman who first reported the phenomenon in 1928.¹³⁷ To obtain a Raman spectrum, a monochromated light source (i.e. laser) is shone on a sample and the intensity of scattered light as a function of frequency is measured.

Raman scattering can be classified as two relative probabilities, Stokes ($\omega_0 - \omega_q$) and Anti-Stokes ($\omega_0 + \omega_q$). In Stokes Raman scattering, the chemical bonds in the material absorb some of the energy thus the molecule is left in an excited state, as illustrated in Fig. 2.21(a). The scattered light is lower in energy (frequency), whereas it is conversely higher in energy (frequency) for Anti-Stokes scattering (Figure 2.21(a)). If the polarizability of the molecule is altered after interacting with the incident photon, the vibration is said to be Raman active, differentiating it from vibrations that are infrared (IR) active whereby the dipole moment must change. The energy change in light is

dependent on bonds and symmetry in the material, allowing Raman signals to act as a characteristic fingerprint of materials.¹³⁸

Raman spectra are typically displayed in terms of Raman shift with units of wavenumbers (inverse wavelength, cm⁻¹) relative to the excitation light (e.g. 532 nm). This Raman shift represents the difference in energy between the incident and emitted photon, thus giving the energy difference of the initial and final vibrational levels of the molecule. The frequency shift of the light depends on the bonds and symmetry in the material, producing a characteristic peak for each Raman active vibrational mode. Fig. 2.21(b) shows a typical Raman spectrum of molecular vibrations with mirrored positive and negative Raman shifts, for Stokes and Anti-Stokes respectively. However, the Anti-Stokes peaks are much lower in intensity than the Stokes under normal conditions due to the Maxwell-Boltzmann distribution law that describes the population of thermally excited vibrational states.¹³⁹ Since both give the same information, it is conventional to measure the Stokes side of the spectrum.¹⁴⁰ The large intensity at 0 cm⁻¹ is where all reflected and Rayleigh scattered light is detected. An application of the Anti-Stokes signal is coherent Anti-Stokes Raman spectroscopy (CARS), which is used to enhance the naturally weak spontaneous Raman signal.¹⁴¹

2.7.1 Classical Raman Theory

Though quantum mechanical Raman theory is necessary to fully describe the Raman phenomenon, classical Raman theory can describe most Raman effects. The following is adapted from 'Introductory Raman Spectroscopy' by J. R. Ferraro, K. Nakamoto and C. W. Brown.¹³⁸

In classical Raman theory, a molecular dipole moment $\vec{\mu}$ in the system, induced by an electric field \vec{E} , can be defined as:

$$\vec{\mu} = \alpha \vec{E} \tag{2.22}$$



Figure 2.21: (a) Jabłoński schematic of the excitation to, and relaxation from, an excited energy state for Rayleigh, Stokes and Anti-Stokes scattering. **(b)** Typical Raman spectrum showing the three scattering types. (Figure courtesy of Dr. John McManus)

where α is the proportionality constant known as polarisability, i.e. how much the electron density is perturbed by the electric field. The electric field strength of the electromagnetic wave \vec{E} (i.e. laser beam) fluctuates with time *t*:

$$\vec{E} = \vec{E}_0 \cos(\omega_0 t) \tag{2.23}$$

where \vec{E}_0 is the vibrational amplitude and ω_0 is the frequency of the laser. If the molecule is vibrating with a frequency ω_q , the nuclear displacement q is written as:

$$q = q_0 \cos(\omega_q t) \tag{2.24}$$

where q_0 is the vibrational amplitude. The polarisability is influenced by the nuclear motion due to the geometry of the nucleus constantly changing to minimise the energy of the system. This change in polarisability can be described using a Taylor series expansion:

$$\alpha = \alpha_0 + \left(\frac{\partial \alpha}{\partial q}\right)_{q_0} q_0 + \dots \tag{2.25}$$

where α_0 is the polarisability at the equilibrium position and $(\partial \alpha / \partial q)_{q_0}$ is the rate of change of α with respect to the change in q at the equilibrium position. Combining the previous four equations, we obtain:

$$\mu(t) = \underbrace{\alpha_{0}\vec{E}_{0}\cos(\omega_{0}t)}_{\text{Rayleigh}} + \underbrace{\frac{1}{2}\left(\frac{\partial\alpha}{\partial q}\right)_{q_{0}}q_{0}\vec{E}_{0}\cos[(\omega_{0}-\omega_{q})t]}_{\text{Stokes}} + \underbrace{\frac{1}{2}\left(\frac{\partial\alpha}{\partial q}\right)_{q_{0}}q_{0}\vec{E}_{0}\cos[(\omega_{0}+\omega_{q})t]}_{\text{Anti-Stokes}}$$

$$(2.26)$$

In this equation, the first term represents Rayleigh scattering, the second and third terms representing Stokes scattering, where incident light re-emitted is red-shifted, and Anti-Stokes scattering, where the emitted light is blue-shifted. If $(\partial \alpha / \partial q)_{q_0} = 0$, the vibration is not Raman-active.

2.7.2 RAMAN OF TMDs

For TMDs, Raman spectroscopy can give information on chemical species/elemental composition, crystal structure, phase, defects, strain, doping and even film thickness/number of layers.^{76,142,143} The Raman spectra of TMDs are generally characterised by two main peaks, corresponding to the in-plane and out-of-plane motion of the atoms. These key modes for 2H and 1T configurations are summarised in Fig. 2.22(a) and (b). Additional modes can be observed in the low-frequency ($< 50cm^{-1}$) region of the Raman spectrum of TMDs. These are referred to as the shear mode (SM) and layer-breathing mode (LBM).¹⁴⁴ These low-frequency modes occur due to relative motions of the planes themselves, either in-plane or out-of-plane to the atomic layers, as illustrated in Fig. 2.22(c). As such, these modes only occur for multilayers, i.e. ≥ 2 layers (L), and can be used to further characterise 2D materials e.g. layer number and stacking sequence.^{145,146}

Single layer MoS_2 and WS_2 both show D_{3b} symmetry.^{143,147} The unit cell consists of one metal atom (Mo/W) sandwiched between two sulfur atoms. For a MoS₂ monolayer, the modes can be



Figure 2.22: Typical in-plane (E_g/E_{2g}^1) and out-of-plane (A_{1g}) Raman active vibrational modes of TMDs of (a) 2H-MoS₂/WS₂ and (b) 1T-PtSe₂. (c) Overview of the layer breathing mode (LBM) and shearing mode (SM) of TMDs in low-frequency Raman spectroscopy.

decomposed into the following irreducible representations: ¹⁴⁸

$$\Gamma_{1L}^{M_0S_2} = A_{1g} + E_{1g} + 2A_{2u} + 2E_{2g}^1 \tag{2.27}$$

Though the A_{1g} , E_{1g} and $E_{2g}^{1}^{\ddagger}$ symmetries are all Raman active modes, the E_{1g} mode requires a scattering geometry with a z component which cannot be significantly observed experimentally in the 2D monolayer. Pristine CVD-grown monolayer MoS₂ typically has Raman peaks observed at ~ 384 cm⁻¹ and ~ 405 cm⁻¹, corresponding to in-plane (E_{2g}^{1}) and out-of-plane (A_{1g}) vibrational modes respectively, when excited at 532 nm. ^{145,149,150}

The WS₂ Raman spectrum can be similarly characterised *via* the first order modes. Pristine WS₂ exhibits a characteristic peak at 352 cm⁻¹, which, when excited at 532 nm, is a combination of several different modes. Of these, the most intense are the in-plane vibrational mode (E_{2g}^1) and the resonantly excited 2LA(M) phonon mode.^{145,151}

As $PtSe_2$ exhibits a D_{3d} point-group symmetry, it produces the following irreducible modes at the

 $^{^{\}ddagger}$ The true labelling for ML MoS $_2$ symmetries are A'_1, E'' and E'. For simplicity, the labelling convention for bulk MoS $_2$ is used. $^{\rm I43}$

centre of the Brillouin zone:

$$\Gamma_{1L}^{PtSe_2} = A_{1q} + E_q + 2A_{2u} + 2E_u \tag{2.28}$$

 A_{1g} (out-of-plane) and E_g (in-plane) are the two Raman-active modes, as shown in Figure 2.22(b). The E_g and A_{1g} modes are generally observed at ~ 175 cm⁻¹ and ~ 205 cm⁻¹ respectively for the ML when excited at 532 nm. As layer number is increased, the E_g mode is gradually red-shifted to lower wavenumbers.^{76,146} In addition, the A_{1g}/E_g intensity ratio gradually increases with increasing layer thickness.^{76,146} The less intense feature at ~ 230cm⁻¹ is assigned to an overlap between the A_{2u} and E_u modes, which are longitudinal optical (LO) modes involving the out-of-plane and in-plane motions of Pt and Se atoms respectively.⁷⁶ This contribution is most evident with close-to-ML thickness and decreases in relative intensity with increasing flake thickness.

2.8 Photoluminescence

This section is adapted from the books edited by C. S. S. Kumar and by D. L. Andrews, G. D. Scholes and G. P. Wiederrecht. .^{152,153}

Luminescence is a phenomenon involving the spontaneous emission of light radiation from a substance or material following the absorption of energy into an excited state. Photoluminescence (PL) is where the energy absorbed is in the form of photons, leading to the electrons in the ground state being excited to a higher energy state and after some energy loss (relaxation), the excited electrons return to ground state along with the emission of a photon.

PL is a powerful emission spectroscopy technique for characterising and studying the properties of nanomaterials, due to its relative simplicity. Again, similar to Raman, it is a non-invasive and non-destructive technique. PL spectroscopy involves the excitation of a sample with a

monochromatic light source, typically a laser, and the detection of the emitted light. A PL spectrum is obtained by scanning and measuring the emitted light.

Room temperature (RT) PL is often used as an indicator of the crystal quality of the material, but this does not take into account the possible presence of non-radiative defects and the fact that the ratio of near-band edge emission to defect emission is strongly dependent on the measurement conditions.

PL can provide much information on the optical properties of semiconductors, such as both intrinsic and defect-related recombination process, as well as giving an overview of the SC material's optical quality. In a bulk SC, the absorption of a photon leads to excitation of an electron from the VB to the CB. The charge carriers are separated by distances than encompass several molecules or ions. This separation, along with the high dielectric constant of the material, makes their binding energy relatively small. The minimum amount of energy required to form the charge carriers is known as the SC's band-gap energy (E_g). The absorption of photons with energy similar to that of the band gap, $h\nu \ge E_g$, leads to an optical transition producing a hole in the VB and an electron in the CB. Absorption of photons with energy greater than E_g produces excitations above the CB edge. The resulting electrons can lose the excess energy through non-radiative processes. Therefore PL spectroscopy is a direct method for determining the bandgap in TMDs.



Figure 2.23: Simplified band structure of bulk MoS₂. The black line is the conduction band minimum (CBM). The two orange lines show the highest valence bands split by the interlayer interactions. The two arrows are the direct-gap transitions showing the A exciton (red) and B exciton (blue).¹⁵⁴

Most Mo- and W- based TMDs exhibit a transition from an indirect to a direct bandgap SC with decreasing layer thickness down to the monolayer which can explain the observed jump in ML photoluminescence efficiency, ⁶⁶ where the PL quantum yields in ML MoS₂ increase by up to a factor of $\sim 10^4$ from the bulk. ^{18,64} This results in ML MoS₂ on Si/SiO₂ exhibiting strong PL at RT from the A exciton[§] which is the most prominent peak at ~ 1.85 eV (670 nm). ^{149,150,156} ML MoS₂ has a second weaker component in its PL, referred to as the B exciton, at ~ 2 eV. ¹⁵⁴ The two components are known to arise from direct-gap optical transitions between the maxima of split valence bands (ν 1, ν 2) and the conduction band minimum (CBM), all located at the K-point of the Brillouin zone, as n in Fig. 2.23. ^{64,154} Pristine ML WS₂ also exhibits a strong PL response with only one direct electronic transition at ~ 1.95 eV,⁷¹ which is up to three orders of magnitude stronger than that of multilayers.⁶⁹

One issue that can affect spectra are the measurement conditions such as temperature and excitation energy.¹⁵⁷ The PL of TMDs in particular is very sensitive to changes in structure and environment, with the crystal quality influencing the intensity, peak position and the FWHM of the PL peak.¹⁵⁶ Similar to Raman, MoS₂ PL is known to be sensitive to temperature,^{158,159} strain,¹⁶⁰⁻¹⁶² dielectric,¹⁶³ defects,¹⁶⁴⁻¹⁶⁶ substrate^{167,168} and dopants.^{169,170} Mouri *et al.* demonstrated that p-type doping with high electron affinity chemical dopants seems to enhance PL intensity, while n-type doping tends to reduce it.¹⁷¹

[§]Excitons, according to the Wannier-Mott model, are quasiparticles that are a combination of an electron and a positive hole bound by Coulomb interaction. A characteristic feature of excitons is that their formation (e.g. at optical excitation) does not lead to the seperation of carriers, thus the excitons are electrically neutral excitations.¹⁵⁵
Certain people - men, of course - discouraged me, saying science was not a good career for women. That pushed me even more to persevere.

Françoise Barré-Sinoussi, virologist & discoverer of HIV

3 Experimental Methods

IN THIS CHAPTER, A GENERAL OVERVIEW and details of the various pieces of equipment and techniques used to synthesise, fabricate and characterise the materials explored in this thesis are provided. The theory behind these techniques is largely contained in Chapter 2, except in some cases where it was deemed more suitable to be included here.

3.1 GENERAL MATERIALS

All materials synthesised and exfoliated in this work were grown/transferred to SiO_2/Si substrates purchased from University Wafers and Dasom RMS. These substrates consisted of 500 μ m thick Si (100) with a top layer of 300 nm SiO₂ obtained by dry thermal oxidation. The Si substrates are cleaned before every CVD run and post-mechanical exfoliation transfer by rinsing the substrates in acetone followed by isopropanol (IPA). The substrates are then dried using nitrogen gas (N₂). This cleaning process removes the majority of large surface contaminants.

All chemicals were ordered from Sigma-Aldrich/Merck unless specified otherwise and were used as received. Metal targets for metal film deposition were obtained from HMW-Hauner. All gases were supplied by BOC gases.

TMD crystals for mechanical exfoliation (ME) were purchased from HQ Graphene. Nitto Denka BT-150E-CM and Scotch Magic tape were used during the mechanical exfoliation process.

3.2 MATERIAL SYNTHESIS

The fabrication of materials used in this thesis can be experimentally divided into two categories: mechanical exfoliation (ME) and vapour phase deposition (VPD), which can be further subdivided into thermally assisted conversion (TAC) and chemical vapour deposition (CVD).

3.2.1 VAPOUR PHASE DEPOSITION

Both VPD methods were performed in quartz tube furnaces, with a separate dedicated furnace used for each chalcogen examined to minimise cross-contamination (see Fig. 3.1). Both use qualitatively similar process, with the primary difference being the chalcogen heating element. The sulfur heater (low-temperature zone, see left image in Fig. 3.1) was an array of halogen lamps, while the other lowtemperature zone had its own dedicated selenium furnace (see right image in Fig. 3.1). Throughout this work, solid phase chalcogen precursors were used, consisting of either sulfur powder or selenium pellets (< 5 mm, > 99.99 % purity). The melting temperatures of the chalcogens are well below the reaction temperatures and using a lower-temperature zone for the vapourisation of the chalcogen prevents excessive amounts of chalcogen in the furnace. Two distinct types of transition metal (TM) sources were used in the CVD synthesis; a thin layer of TM oxide eg. MoO₃ or a sputter deposited TM film on a SiO₂/Si substrate.



Figure 3.1: Photos of sulfur (left) and selenium (right) two-zone furnaces used in this work for both TAC and CVD processes.

3.2.2 METAL DEPOSITION

For thermally assisted conversion (TAC) of TM (Mo, W, Pt) to TMD, the transition metal thin films were deposited by physical vapour deposition (PVD) onto Si/SiO₂ substrates. Three methods of PVD were used to deposit a controlled thickness of TM; argon-ion sputtering, electron beam evaporation and molecular beam epitaxy (MBE).

For argon-ion sputtering, the sputter coater used was a Gatan Precision Etching and Coating System (PECS) which involves a beam of high energy Ar ions striking a desired metal target, a MaTeck Pt

target in this case, thus sputtering onto the substrate below. The chamber pressure was $\sim 5 \times 10^{-5}$ mbar during the deposition, with a deposition rate between 0.3 to 1 Å/s. While this method is relatively straightforward, the impacting high energy metal atoms can damage the sample surface and any underlying films. These embedded metal atoms may result in gate leakage. In addition, the tool has limited throughput due to $\sim 2 \times 2$ cm sample size limitation.

For e-beam evaporation of TM, a Temescal FC-2000 Evaporation System was used, whereby a high energy beam of electrons bombards a metal target, heating the target until the metal atoms evaporate thus depositing a thin film on the substrate. E-beam evaporation can offer good thickness control and the material flux is a less energetic and therefore a less damaging technique than sputtering. In addition, molecular beam epitaxy (MBE) was used, whereby thin films are grown with precise control over properties such as thickness, stoichiometry and crystal phase. The chosen materials are evaporated in ultra high vacuum (UHV) and the resultant evaporant flux impinges on a substrate,

typically a single crystal held at high temperature. Growth of Pt thin films were carried out in a DCA M600 MBE system with a base pressure of 5×10^{-10} Torr on a variety of Al₂O₃ growth substrates; c-plane sapphire, r-plane sapphire and amorphous alumina. The substrates were annealed under vacuum for \sim 2 hours prior to deposition. Pt was deposited at 600 °C.

All three tools use a quartz crystal microbalance to monitor deposition rate and film thickness.

3.2.3 THERMALLY ASSISTED CONVERSION (TAC)

Once appropriate thicknesses of the desired TM are deposited, the TAC process, as described previously by Gatensby *et al.*⁹⁴ and O'Brien *et al.*⁷⁶, can be utilised to transform the metal into the desired TMD. Though an extremely versatile fabrication technique applicable to many TMDs, the only TMD TAC film examined in this thesis was that of PtSe₂.

Once Pt was deposited, the films^{*} of desired thickness were placed into the centre of a custom-built quartz-tube furnace (see right image in Fig. 3.1) with two heating zones. The tube was then evacuated and the samples were heated to 450 °C under 150 sccm forming gas (9:1 Ar:H₂) flow in the high-temperature zone. A selenium (Se) precursor (solid pellets < 5 mm) was heated independently upstream in the furnace in the second (low-temp) zone (Sigma Aldrich, $\geq 99.99\%$), to ~250 °C, the Se-vapour was carried downstream to the Pt films for a duration of 2 hr to ensure selenisation of the surface. After a 30 min post-anneal at 450 °C in 150 sccm Ar, the furnace was subsequently cooled to room temperature. The forming gas helps to reduce surface oxidation of the metal film.¹⁷² For the growth of selenide TMD films, the presence of H₂ is especially necessary as the selenium is not reactive enough on this temperature/pressure. Se combines with H₂ to form H₂Se which is much more reactive with the transition metal film.¹⁷³ To minimise health risk from exposure to the selenium used in the growth of Se-TMDs, a solid Se precursor was utilised. In addition, a sealed ventilation box was placed over the furnace to prevent Se contamination from entering the lab atmosphere.

3.2.4 CHEMICAL VAPOUR DEPOSITION (CVD)

The CVD synthesis method here was previously developed by O'Brien *et al.*⁹³ and was primarily used to synthesise monolayer (ML) TMDs in the dedicated sulfur furnace, such as MoS₂ and WS₂. This technique can also be applied to selenide TMDs. The notable difference between the CVD and TAC processes is the TM precursor and growth substrate geometry. In CVD, all precursors are evaporated and the materials grown are high in crystallinity, whereas TAC produces polycrystalline thin films.

^{*}From atomic force microscope (AFM) measurements of the thicknesses of Pt layers before and after selenisation, it has been reported that the initial Pt thickness expands approximately four times after post-selenisation.⁹⁶ Henceforth, TAC film thicknesses will be referred to by their starting Pt thickness, unless otherwise stated.



Figure 3.2: Schematic of the two-zone quartz furnace systems used for TMD synthesis in this thesis. Image courtesy of Dr. Conor Cullen.

For the CVD synthesis of TMDs, at least two precursors are necessary; one containing the metal and another containing the chalcogen. The options in chalcogen precursors are usually straightforward, with either H_2S/H_2Se or the vapourised chalcogen being used. Elemental selenium is usually favoured over H_2Se due to its lower toxicity.

The CVD method employs a microreactor geometry, where the growth substrate is brought into close proximity with the seed substrate containing the metal precursor (see Fig. 3.3). For MoS₂, MoO₃ dispersed in solution was dropcasted onto a SiO₂ precursor substrate. Once the solvent is evaporated, a clean growth substrate was then placed face-down on the MoO₃ precursor substrate. Similarly, for WS₂, a dropcasted solution of Bi₂(WO₄)₃ dissolved in NH₃.H₂O (solvent baked off at 100 °C) was used in the microreactor. This stack/sandwich geometry allows allows a sufficient amount of sulfur vapour to intercalate between the wafers and react with the vaporised Mo precursor to generate MoS₂ monolayers which then deposit on the growth substrate.



Figure 3.3: Schematic of the furnace setup used for CVD TMD synthesis in this thesis. On the left is the upstream low temperature zone where the chalcogen is evaporated (yellow) and carried downstream to the high temperature zone, on the right. A detailed close-up is shown of the microreactor with the bottom substrate as the seed layer containing the metal precursor (blue) and the resulting TMD growth on the top substrate. Image courtesy of Dr. John McManus.

3.2.5 MECHANICAL EXFOLIATION

Mechanical exfoliation (ME) and transfer are the most successful techniques to obtain high-quality single or few-layer crystalline materials from their native bulk structures, particularly for graphene and TMDs.⁸² For this work, a custom mechanical transfer stage was built (see Fig. 3.4), based on the experimental setup outlined by Castellanos-Gomez *et al.*¹⁷⁴ To prepare samples with ME flakes on the substrate, PDMS stamps (Gel-Pak-4) were used in transferring the thinned flakes from Nitto/Scotch tape to the desired substrate using a transfer stage. Once the flakes had been transferred from the stamp to the substrate, the sample was annealed under Ar at 150 °C to ensure good adhesion, especially for spin-coating purposes.



Figure 3.4: Left: photo of custom-built mechanical transfer apparatus. Right: optical microscope images of ME PtSe₂ flakes transferred onto a SiO₂/Si substrate using this setup. Darker areas of flakes indicate very few layers.

3.3 LITHOGRAPHIC TECHNIQUES

3.3.1 Photo & Electron Beam Lithography

To create samples that were suitable for electrical characterisation, lithographic techniques were necessary to create channel devices. Two approaches were used: UV photolithography (masked and maskless) and electron beam lithography (EBL).

UV lithography is the standard technique for integrated circuit (IC) fabrication. It involves UV light shining through a patterned mask onto a substrate spin-coated with a light-sensitive polymer called resist. The excess photoresist is removed by immersing the substrate in developer followed by rinsing in DI water. Positive photoresist becomes soluble in the developer whereas negative photoresist becomes insoluble after exposure to UV. The lithography tools used in this work were an OAI mask aligner situated in a Class 100 cleanroom and a maskless lithography setup, Intelligent micro patterning (IMP). The IMP operates by projecting a monochrome image onto the substrate, which can be aligned using the in-built microscope. This generally involved two steps in fabrication of TAC devices: first - markers and channels, and second - contacts (post-selenisation). For ME flakes, only the contacting step was required.

EBL is a powerful technique for creating patterns and nanostructures that are too small to fabricate using conventional photolithography, achieving resolutions up to a few nm. A highly focused beam of electrons is rastered over a sample, again spin coated with (electron sensitive) resist, to write out the desired pattern. The e-beam induces a change in the molecular structure of the resist, altering its solubility. This results in cross-linking in the case of negative tone resist and chain scission in the case of positive tone resist. Like photolithography, the substrate is immersed in developer to selectively dissolve either the exposed/unexposed areas of resist. Three steps were generally involved in



Figure 3.5: Photos of IMP tool used for maskless UV lithography.

fabricating the channel devices using EBL; markers, channels and then contacts post-selenisation. A Karl Zeiss Supra 40 field emission scanning electron microscope (FE-SEM)[†] and Raith Quantum software beam blanker and software were used. The samples were first spin-coated with a layer of e-beam sensitive resist, PMMA A6. The exposure areas were removed with a developer solution (IPA:MIBK 4:1).

For all techniques of device fabrication, positive resist was used to allow precise metal deposition. Once, the metal was deposited, the remaining resist was removed in lift-off step in acetone, leaving clearly defined channels and/or contacts.

[†]Scanning electron microscopy (SEM) is described in further detail in Section 3.4.5.

3.4 Microscopic Techniques

3.4.1 Atomic Force Microscopy (AFM)

For this project, the AFMs used were an Asylum Research Cypher and primarily a Bruker Multimode 8. The mode used to perform most topographical measurements was ScanAsyst-Air, a PeakForce tapping mode, similar to tapping mode but with the gain and setpoint constantly being re-adjusted automatically through a sophisticated real-time feedback loop, allowing optimal data to be collected whilst minimising noise. Typical probes used in topography measurements were ScanAsyst-Air (silicon nitride, k = 0.4 N/m, $f_0 = 75$ kHz, Bruker) and SSS-NCH (etched silicon, k =42 N/m, $f_0 = 330$ kHz, Nanosensors). The subsequent images were analysed and their topographical profiles extracted using the Gwyddion software.

3.4.2 NANOSHAVING/NANOMACHINING

Tips used to nanomachine/nanoshave were wear-resistant diamond with broad cone angles (NM-RC, k = 350 N/m, $f_0 = 750$ kHz, Adama Innovations Ltd., see Fig. 2.7(b)). Prior to any AFM manipulation, the tip was underwent a contact mode force calibration to determine the deflection sensitivity of the cantilever. This involved performing a thermal tune followed by multiple ramps (i.e. force-distance curves) on a sapphire substrate (SAPPHIRE-12M, Bruker). From this, the inverse value of the AFM's optical lever sensitivity (InvOLS, nm/V) was measured using the force curve slope on retraction, as previously outlined in Section 2.1.5. This process also gives a more accurate value for the spring constant, k.



Figure 3.6: Photos of the Bruker Multimode 8 AFM, showing several components and aspects of the system.

Once the shaving/machining tip was calibrated, the key operating parameters that determined the quality of nanomachining are:

- 1. Deflection setpoint
- 2. Scan speed.

2D TMD materials were nanomachined/nanoshaved at various scan speeds ($10-30 \mu m/s$) and setpoints (0.01-0.5 V) to optimise parameters in contact mode using a Bruker Multimode 8.

3.4.3 Kelvin Probe Force Microscopy (KPFM)

As mentioned in Section 2.1.7, a bias can be applied either through the tip or the sample in KPFM. In this work, the bias was applied through the sample as the CPD images of higher potential translate to higher work function, and vice versa. For a sample-biased system, the work functions, φ , can be determined from the CPD as such:⁴²

$$CPD = V_{probe} = \frac{\varphi_{sample} - \varphi_{probe}}{|e|}$$

$$\varphi_{probe} = \varphi_{Au} - |e|CPD_{Au}$$

$$\varphi_{sample} = \varphi_{probe} + |e|CPD_{sample}$$
(3.1)

Where *e* is the elementary charge.

PeakForce KPFM (PF-KPFM) was the primary KPFM mode used in this work on the Bruker Multimode 8. As mentioned in Section 2.1.7, PF-KPFM is a two pass mode that combines Bruker's proprietary PeakForce tapping mode and standard FM-KPFM. Like other electrical AFM modes, samples were electrically connected directly to the chuck through silver paint (Agar Scientific) contact from the sample to its magnetic puck.



Figure 3.7: PeakForce Kelvin probe force microscopy calibration under different biases. (a) PF-KPFM topography and (b) corresponding surface potential map while tip biased. (c) Schematic and optical of KPFM calibration Au-Si-Al sample. (d) Contact potential difference (CPD) line profiles of tip biased (b, black) and sample biased (c, red) with corresponding material regions labelled. (e) PF-KPFM topography and (f) corresponding surface potential map while sample biased.

For all KPFM measurements throughout this work, three different electrically conductive probes were used:

- SCM-PIT-V2, antimony-doped Si, k = 3 N/m, $f_0 = 75 \text{ kHz}$, Bruker
- PFQNE-Al, silicon nitride, k = 0.8 N/m, $f_0 = 300 \text{ kHz}$, Bruker
- AD-2.8-AS, boron-doped single crystal diamond, k = 2.8 N/m, $f_0 = 75$ kHz, Adama Innovations Ltd.

To calibrate the tips, a KPFM calibration sample (PFKPFM-SMPL, Bruker) was used (Fig. 3.7(c)), which is patterned with Au, Si, Al strips. Both the Au and Al films deposited are ~ 50 nm in thickness on n-doped silicon substrate. A staircase potential profile is expected across the 3 different materials (see Fig. 3.7(d)). KPFM calibration measurements under sample bias were compiled for statistical purposes to produce Fig. 3.8(e). φ_{probe} was calculated using the work function of the Au strip φ_{Au} which is assumed to be 5.12 eV based on literature values. ${}^{175-177} \varphi_{Au}$ is used in the probe calculations rather than φ_{Al} as aluminium forms an oxide in ambient conditions, thus gold is more stable. From this, φ_{probe} was determined to be 5.31 \pm 0.27 eV for SCM-PIT-V2, 4.69 \pm 0.22 eV for PFQNE-Al and 5.50 \pm 0.20 eV for AD-2.8-AS. In addition, the lift heights were optimised for each tip type prior to imaging. Optimal heights were determined to be 85 nm for SCM-PIT-V2, 40 nm for PFQNE-Al and 75 nm for AD-2.8-AS.



Figure 3.8: (a) and (c) show CPD maps under tip (a) and sample (c) bias of CVD-grown monolayer MoS₂ after nanoshaving, with their respective topographical images (b) and (d). (e) Calibrated work function ranges for probes using throughout this work.

3.4.4 Conductive AFM

AFM conductivity measurements were carried out using the conductive PeakForce tapping mode, PF-TUNA, on the Bruker Multimode 8. Like other electrical AFM modes, samples were electrically connected directly to the chuck through silver paint (Agar Scientific) contact from the sample to its magnetic puck. The sample was biased at 1 V during measurements. The tips used were AD-40-SS (k = 40 N/m, $f_0 = 200$ kHz, Adama Innovations Ltd.) – super sharp (tip radius < 5 nm) conductive single crystal diamond probes. To allow conductivity in the tip, the diamond coating is highly doped with boron leading to a macroscopic resistivity of $0.003-0.005 \Omega$ cm.

3.4.5 Scanning Electron Microscopy

The scanning electron microscope, was invented by Manfred von Ardenne in 1937.¹⁷⁸ It was the first microscope to achieve high magnification by scanning a very small raster with a finely focused beam of electrons. Although there had been some crude models before his, von Ardenne was the first to succeed in eliminating the chromatic aberration that had plagued previous electron microscopes. The SEM uses a focused beam of high-energy electrons to generate signals as their energy is dissipated at the surface of specimens. The signals that derive from electron-sample interactions contain information on topography and composition of a sample's surface.

Generally in SEM, data is collected over a selected surface area in a raster fashion and an image is generated. However, like the AFM, there are some limitations to the SEM. Firstly, the samples must be solid stable in a vacuum on the order of $10^{-5} - 10^{-6}$ torr. Any 'wet' samples, such as organic materials, decrepitate/outgas at low temperatures. Nevertheless these samples can be examined in a special 'environmental' SEMs at a relatively low vacuum. Also, an electrically conductive coating must be applied to electrically insulating samples for study in conventional SEMs to avoid charging effects, unless the instrument is capable of operation in a low vacuum mode.

The SEMs used in this thesis were Karl Zeiss Supra 40 for EBL and Karl Zeiss Ultra plus for general imaging. The Karl Zeiss Ultra plus was operated at 5 kV accelerating voltage, 30 μ m aperture and a working distance of typically 5 – 7 mm. Primarily, imaging was acquired using the secondary electron detector (SE2).

3.5 Spectroscopic Techniques

3.5.1 RAMAN SPECTROSCOPY

Raman spectroscopy was carried out using a WITec Alpha 300R spectrometer with a 532 nm excitation laser, which was fitted with a Rayshield Coupler to detect Raman lines close to the Rayleigh line at 0 cm^{-1} . Spectra were recorded using a laser power of $< 100 \,\mu W$ for MoS_2 and WS_2 and < 300 μW for PtSe_2 in order to minimise sample heating. A 100x objective with a numerical aperture (NA) of 0.95 and a spectral grating with 1800 lines/mm were used for all Raman spectra. Maps were generated by taking 4 spectra per µm in both x and y directions over large areas. The resulting data was analysed and the spectra were subsequently graphed using a combination of WITec Project FIVE and Origin Pro software.



Figure 3.9: Photograph of WITec Alpha 300R scanning Raman and PL spectrometer used in this work.

3.5.2 Photoluminescence

In this work, all PL measurements and spectra were recorded using a WITec Alpha 300R at RT with a 532 nm laser as the excitation source. A 100x objective with a numerical aperture (NA) of 0.95 and

a spectral grating with 600 lines/mm were used for all PL measurements. Spectra were recorded using a laser power of $< 100 \,\mu\text{W}$ for MoS₂ and WS₂. Cold temperature PL measurements were carried out at 77 K. Maps were generated by taking 4 spectra per μ m in both x and y directions over large areas. The resulting data was analysed and the spectra were subsequently graphed using a combination of WITec Project FIVE and Origin Pro software.

3.6 Electrical Characterisation

The PtSe₂ TAC devices were fabricated using shadow-masked UVL and EBL to pattern the substrate in the desired geometry. Pt was deposited using e-beam evaporation and subsequently selenised using the TAC method. Contacts were deposited *via* e-beam evaporation, comprised of a 5 nm Ti (adhesion layer) with 45 nm Au on top. The ME PtSe₂ flake devices were fabricated using maskless UVL. E-beam evaporation was employed to deposit contacts consisting of 5:45 nm Ti/Au. The electrical characterisation of the TMD devices was carried out in ambient conditions using a Karl Suss probe station connected to a Keithley 2612A source meter. The voltage was applied through needle-probes, contacting Au electrodes deposited on the TMDs. The measured current was collected and plotted automatically using Labview and Origin Pro software to give current-voltage (I-V) curves.

All sorts of things can happen when you're open to new ideas and playing around with things.

Stephanie Kwolek, chemist & inventor of Kevlar

4

Patterning 2D Surfaces by Nanoshaving

4.1 INTRODUCTION

SCANNING PROBE MICROSCOPY (SPM) TECHNIQUES have been widely employed in the fabrication of nanostructures on a wide variety of surfaces.^{47–49} One such technique is atomic force microscopy (AFM), a versatile tool that allows users to not only probe a samples' electrical, magnetic

and mechanical properties in a wide range of environments at very high resolution, but also to manipulate materials down to the nanometre scale. ^{179,180} Self-assembled monolayers (SAMs) have proved to be ideal resist materials for such physical manipulation and fabrication. ^{50,104,181–183} Nanoshaving is when an SPM tip is used to ablate a surface layer mechanically or electrochemically with a required normal force. ⁵⁰ It differs from nanomachining as it selectively removes adsorbed layers (adlayers) rather than penetrating the sample surface. ⁵¹ Mechanical displacement, or nanoshaving of SAMs by an AFM probe is a relatively inexpensive technique, ideal for fundamental studies. ⁵⁰

Despite being highly researched materials for a wide range of applications, to date, there has been little exploration of the nanopatterning of SAMs on 2-dimensional (2D) materials using AFM. The versatile material removal applications of AFM make it an excellent candidate for investigating non-covalently functionalised 2D systems and it offers advantages over established techniques such as bottom-up approaches and focused ion beam (FIB) milling. ^{52,184} The AFM-based approach avoids problems related to ion-injection from FIB, ¹⁸⁵ low resolution (~ 300 nm) of confocal Raman microscopy ¹⁸⁶ and constraints on nanostructure size and shape due to mask limitations in bottom-up approaches. ¹⁸⁷ AFM nanopatterning also enables data to be collected *in situ* during the nanoshaving/nanomachining process, giving insight into the mechanical properties of the material. The mechanical nature of the nanoshaving process is the root of its limitations. The SAM adsorbates (resist) should be selectively removed without substantially damaging the underlying 2D material. Different factors such as sample material, tip speed, applied force, tip angle, sliding direction, humidity and sample orientation affect how a material is removed.⁵³ Furthermore, AFM-based mechanical nanoshaving has the issue of ridge formation or the accumulation of materials around the inscribed features.⁵⁴

Chemical functionalisation of layered 2D materials, such as graphene or transition metal dichalcogenides (TMDs), is essential in modifying their properties and expanding their capabilities

and applications.⁹⁷ Exploiting the functionalisation of TMDs is seen as a route to their widespread application of TMDs in photonics, energy storage and conversion, drug delivery and medical devices.^{98,99} TMD surfaces tend to be rather inert to chemical functionalisation, due to saturation of basal plane chalcogen atoms.^{98,100} Meanwhile, the metal sites sandwiched in between the chalcogen layers are rendered somewhat protected from functionalisation. Fortunately, TMDs have a tendency to be very sticky, as evidenced by the many organic groups that readily physisorb through electrostatic interactions to the electron-rich chalcogen surface atoms.⁹⁸ As has been previously shown for graphene, non-covalent functionalisation of 2D materials is an attractive strategy for altering their surface chemistry without damaging their electrical and mechanical properties.^{188–190}



Figure 4.1: Chemical structures of **(a)** perylene-3,4,9,10-tetracarboxylic dianhydride (PTCDA) and **(b)** N,N'-substituted perylene-3,4:9,10-tetracarboxdiimide, a perylene diimide (PDI) derivative.

Würthner *et al.* outlined how perylene-based dyes have been implemented in many fields of technology due to their excellent physical and optical properties and their tendency towards self-assembly.¹⁹¹ Perylene is a polycyclic aromatic hydrocarbon consisting of two bonded naphthalene molecules and its derivatives range from intense orange to red in colour.¹⁹² Perylene-based π -surfactants can be ideally used as probes as they possess the characteristic

spectroscopic fingerprint features (absorption, emission, Raman) of both the aromatic dispersant and the respective carbon allotrope.¹⁹³ A key primary perylene is perylene-3,4,9,10-tetracarboxylic dianhydride (PTCDA, Fig. 4.1(a)), an organic semiconductor.¹⁹⁴ Peng *et al.* determined an increase of the electroactive surface area and superior conductivity in graphene with PTCDA as a surfactant.¹⁹⁵ However, one fundamental drawback of primary perylene and PTCDA is their intrinsic insolubility in organic solvents and water.¹⁹³

surfactant Another perylene of interest in systems N,N'-substituted is perylene-3,4:9,10-tetracarboxdiimide,¹⁹⁶ a derivative of perylene diimide (PDI, Fig. 4.1(b)), which has been frequently used for the dispersion and stabilisation of carbon allotropes in aqueous solutions.¹⁹³ PDI-based molecules exhibit strong absorption through their large aromatic core, which can become attached to 2D layers via van der Waals (vdW) interactions. 197-199 Berner et al. previously investigated the packing density of PDI for the non-covalent functionalisation of single-layer graphene grown via chemical vapour deposition (CVD).¹⁸⁸ Depending on the presence of substrate contamination, the aqueous-phase deposition of PDI resulted in different packing densities of SAMs which were directly observed in scanning tunneling microscopy (STM). Wirtz et al. showed that perylene derivative SAMs can be used to improve adhesion to MoS₂, acting as seeding layers for atomic layer deposition (ALD) oxides 197 while Kim et al. developed this methodology for the passivation of MoS₂ monolayer channels in FETs, significantly improving performance.⁹⁹ Additionally, Abellán et al. reported the formation of a protective layer with an electron-deficient PDI derivative which conferred black phosphorus flakes with considerable stabilisation against oxygen degradation.^{200,201}

For any electronic material, a key parameter is the work function (φ), which is defined as the difference in energy level at vacuum and the Fermi energy ($WF = E_{vac} - E_F$)²⁰² and interprets how that material interacts when interfaced with other materials/elements. The work function also controls the charge transfer and transport across the interface, thus it is important to study the

electrostatic potentials of TMD/organic heterostructures and the effect of the organic layer in modulating the TMD work function.²⁰³ Kelvin probe force microscopy (KPFM) has been used to characterise the charge distribution of various materials by measuring the surface potential/contact potential difference (CPD) between the AFM tip and sample surface.⁴¹ It has been used to measure the contact resistance and potential drops in functionalised graphene²⁰⁴ and bilayer MoS2²⁰⁵ between contacts under external bias conditions, as well as organic thin film transistors.²⁰⁶ By engineering specific surface dipoles, the work function can be intentionally modified. Therefore, KPFM is a useful technique for characterising the charge transfer at the interface between 2D TMDs and non-covalent adlayers. One challenge with KPFM can be that adsorbed species from ambient air, such as water and carbon, change not only the topography but also the surface potential through charge transfer doping. Kahn discusses how exposure of the semiconductor surface to ambient air or other contaminants can affect work function, electron affinity and ionisation energy.²⁰²

When predicting the properties of interfaces between divergent materials, like organic SAM-inorganic 2D heterojunctions, it is helpful to examine these by way of an energy level diagram. For the SAM-2D case, the relevant energy levels are the highest occupied molecular orbital (HOMO) and lowest unoccupied molecular orbital (LUMO) vs. the 2D valence band maximum (VBM) and conduction band minimum (CBM).²⁰⁷ The likely flow of charges in the ground and excited states can be predicted using this energy level alignment. This approach can also be used to identify potential SAM-2D material combinations for photovoltaic cells, photodetectors, diodes and other electronic and optical devices.

Dipoles and charges in SAM molecules can change the electronic structure of the 2D material in a similar manner to a gate voltage in a transistor, moving the 2D Fermi level (E_F) up or down in energy.²⁰⁷ Many commonly used organic SCs, such as planar phthalocyanines and perylenes, are neutrally charged with either small dipoles or no net dipole moment. In these systems, the movement of charge carriers is determined by hybridisation and ground state-charge transfer.²⁰⁷

Habib *et al.* reported that in PTCDA/MoS₂ heterojunctions, hybridisation mixes the S p_z and Mo d_{z^2} orbitals near the CBM and the conjugated C p_z orbitals.²⁰⁸ This hybridisation results in the narrowing of the MoS₂ bandgap and increases the density of states (DOS) near the band edges, resulting in a large intensity increase and a red shift in the PL spectrum. In addition, a net charge transfer occurs due to the enhanced DOS near the conduction band minimum (CBM) and overlapping of the excitonic energy levels in the heterostructure (see Fig. 4.2(b)). A similar type-II band alignment was also predicted by Obaidulla *et al.* (see Fig. 4.2(a)).²⁰⁹



Figure 4.2: Previously reported Type-II band alignment illustrations for organic-inorganic heterostructures of PTCDA on MoS₂ by (a) Obaidulla *et al.*²⁰⁹ and (b) Habib *et al.*²⁰⁸

This chapter investigates the work function change associated with the non-covalent functionalisation of monolayer TMDs with PDI SAMs, as measured by KPFM. The reversibility of such functionalisation is then examined through the approach of nanoshaving. Another perylene, PTCDA, which is chemically distinct from PDI, was deposited on the shaved areas to demonstrate non-covalent co-functionalisation of TMDs. A grating pattern was fabricated using nanoshaving as a lithographic technique. A diffraction grating was chosen as it is one of the most essential optical

components used in many academic and industrial areas²¹⁰ and is relatively simple in structure.

4.2 EXPERIMENTAL METHODS

4.2.1 CVD of TMDs

MoS₂ and WS₂ samples were grown in a micro-cavity in a two-zone CVD furnace at 700 °C and 900 °C respectively, using the method outlined by O'Brien *et al.* ⁹³ The MoS₂ samples were grown directly on SiO₂/Si substrates (300 nm thick oxide layer, highly p-doped Si) with MoO₃ dropcasted from solution and sulfur as solid precursors for MoS₂. A dropcasted solution of Bi₂(WO₄)₃ dissolved in NH₃.H₂O (solvent baked off at 100 °C) and sulfur served as precursors for WS₂.

4.2.2 Surface Functionalisation of TMDs by SAMs

Directly after TMD growth, perylene diimide (PDI, Fig. 4.1(b)) dissolved in aqueous buffer solution (1 mM) was deposited by dropcasting on the TMD for the functionalisation step. Post-shaving of PDI on MoS_2 , PTCDA (Fig. 4.1(a)) dissolved in dimethylformamide (DMF, 1 mM) was deposited on the MoS_2 in the same fashion as PDI.

4.2.3 AFM CHARACTERISATION & MANIPULATION

AFM nanoshaving and KPFM were performed in ambient conditions using a Bruker Multimode 8. All nanoshaving operations were carried out using nanomechanical diamond cone tips (NM-RC, k = 350 N/m, f₀ = 600 kHz, supplied by Adama Innovations Ltd.) in contact mode. The deflection sensitivity of the cantilevers was determined by performing force-distance curves pre-shaving on a sapphire substrate. For KPFM, the tips (SCM-PIT-V₂ and PFQNE-Al, Bruker) were calibrated using a Au-Si-Al calibration sample (PFKPFM-SMPL, Bruker). To minimise possible topographical artefacts, PeakForce KPFM (PF-KPFM), a two-pass (lift) mode, was used, where topographical data was recorded in the first pass. Then the tip was lifted to a user-specified distance above the sample surface (~ 85 nm for SCM-PIT-V₂, ~ 45 nm for PFQNE-Al), measuring the work function in the second pass. Raman and PL spectra were recorded using a WITec Alpha 300R with a 532 nm laser as the excitation source and a laser power of <100 μ W, in order to minimise sample damage. A 100x objective with a numerical aperture (NA) of 0.95 and a spectral grating with 1800 lines/mm were used for all Raman spectra while a spectral grating with 600 lines/mm was used for all PL measurements. Maps were generated by taking 4 spectra per μ m in both x and y directions over large areas.

4.3 RESULTS & DISCUSSION

The feasibility of removing perylene-based SAMs from the surface of monolayer TMDs *via* AFM nanoshaving was examined. CVD-grown MoS₂ samples on 300 nm SiO₂/Si were prepared, along with one CVD-grown WS₂ sample. One sample remained as pristine MoS₂, while others, including WS₂, were functionalised with PDI. One MoS₂/PDI sample was further functionalised with PTCDA post-nanoshaving.

Nanoshaving was carried out in a single pass contact mode on all samples in ambient conditions. Using the cantilever spring constant, k, and Hooke's Law, the contact force is defined as:

$$F_{tip} = kd \tag{4.1}$$

where d = deflection sensitivity (nm/V) \times deflection setpoint (V). It was found that damage of the underlying TMD monolayer began to be observed at loads > 1.5 μ N. Nanoshaving at loads < 1 μ N generally resulted in discontinuous and incomplete removal of material from the TMD surface. The

average load which resulted in the best nanoshaving was 1.25 \pm 0.04 μ N. This was sufficient to remove adsorbate molecules completely without causing damage to the TMD.

KPFM was employed to investigate and compare the shaved TMD, with and without functionalisation. In KPFM, a bias can be applied either through the tip or the sample. In this work, the bias was applied through the sample meaning that the CPD images of higher potential translate to higher work function, and vice versa. The work function is defined as the difference in energy between vacuum and Fermi energy levels ($\varphi = E_{vac} - E_F$).²⁰² For a sample-biased system, the work functions, φ , can be determined from the CPD as such:⁴²

$$CPD = V_{probe} = \frac{\varphi_{sample} - \varphi_{probe}}{|e|}$$

$$\varphi_{probe} = \varphi_{Au} - |e|CPD_{Au}$$

$$\varphi_{sample} = \varphi_{probe} + |e|CPD_{sample}$$
(4.2)

Where *e* is the elementary charge. Measurements were performed using Si cantilevers whose work functions ($\varphi_{probe} = 5.31 \pm 0.27$ eV for SCM-PIT-V2, 4.69 \pm 0.22 eV for PFQNE-Al) were determined by calibration with a Au-Al reference sample.

4.3.1 NANOSHAVING PRISTINE MOS₂

To investigate and characterise the molecular assembly and the effect of nanoshaving on such assemblies, the complementary techniques of KPFM, Raman and PL spectroscopies were used. Typical images acquired during KPFM imaging of MoS_2 prior to nanoshaving are shown in Fig. 4.3(a) and (b), with topography measured on the first pass and surface potential measured on the second pass in a lift mode. For pristine MoS_2 , apart from the accumulation of surface contaminants formed at the shaved area perimeter, there was a negligible decrease in the apparent



Figure 4.3: Nanoshaving of pristine $MoS_2 - (a)$ and (c) show topography of the area of MoS_2 before and after nanoshaving, with their respective contact potential difference (CPD) images (b) and (d). (e) Optical image of the area scanned using Raman and PL spectroscopy (red box) with shaved area indicated by white dashed box. (f) Raman peak intensity map of MoS_2 A_{1g} mode (405 cm⁻¹). (g) PL peak intensity map of MoS_2 A exciton (1.83 eV) with (h) corresponding average spectra of pristine vs. shaved areas.

topographical height of the MoS₂ following nanoshaving (Fig. 4.3(c)). However, there is a significant contrast in the CPD between the nanoshaved and pristine areas of MoS₂, with a slight increase of 0.062 ± 0.062 eV for monolayer MoS₂ and 0.034 ± 0.63 eV for the bilayer in this image. It should be noted that ridge formation of nanoshaved debris typically occurs on the perimeter of the shaving scan area, as seen in Fig. 4.3(c).

Pristine CVD-grown monolayer MoS₂ typically has Raman peaks observed at ~ 384 cm⁻¹ and ~ 405 cm⁻¹, corresponding to in-plane (E_{2g}^1) and out-of-plane (A_{1g}) vibrational modes respectively (see Fig. A.1), when excited at 532 nm.^{145,149} Monolayer MoS₂ on Si/SiO₂ also exhibits strong photoluminescence at room temperature from the A exciton at ~ 1.85 eV (670 nm).¹⁴⁹ From Raman analysis of the area, it is clear in Fig. 4.3(f) there is no significant change in the peak intensity of the A_{1g} mode from nanoshaving. There is also no significant shifting or broadening of the E_{2g}^1 and A_{1g} peaks (see Fig. A.1). Nevertheless, there was a slight decrease in intensity but no shift in position

of the PL signal of the A exciton at 1.83 eV (see Fig. 4.3(g) and (h)). This implies that the MoS_2 is being cleaned of surface contaminants and adsorbates during nanoshaving, rather than removal or destruction of the TMD monolayer.

4.3.2 Nanoshaving PDI on MoS_2



Figure 4.4: Nanoshaving PDI on $MoS_2 - (a)$ topography of PDI functionalised MoS_2 . (b) topography and (c) CPD of nanoshaved PDI/MoS₂ area indicated by white dashed box in (a). (d) corresponding Raman peak intensity map of PDI (1300 cm⁻¹) mode. (e) Raman peak intensity map of $MoS_2 A_{1g}$ (404 cm⁻¹). (f) average Raman spectra of shaved and non-shaved areas, showing characteristic PDI peaks. (g) average PL spectra of shaved and non-shaved areas.

From AFM, the typical thicknesses of CVD-grown monolayer MoS₂ and WS₂ flakes were determined to be 0.7 nm and 0.75 nm respectively. After dropcasting PDI and rinsing with deionised (DI) water and isopropanol (IPA) to remove excess PDI molecules not directly bound to the TMD surface, the samples were examined using AFM/KPFM. The pristine MoS₂ monolayer step height before functionalisation is consistent with previously reported values.^{99,102,211} Raman analysis shows that the PDI appears to self-assemble preferentially to the TMD surface rather than SiO₂ (see Fig. 4.4(d) and Fig. 4.10(c)), as previously observed in other studies.^{102,197} The average

combined step height of monolayer MoS_2 functionalised with PDI SAM was 2 nm, indicating a thickness of ~ 1.3 nm for the PDI SAM. This compares well to previously reported values.¹⁰²

Fig. 4.4(a) shows the topography of a typical MoS₂ flake with a PDI overlayer. Nanoshaving PDI on MoS₂ resulted in a reduction of step height by ~ 1.2 nm (the thickness of a PDI SAM) demonstrating nanoshaving's applicability for lithography. Fig. 4.4(c) shows significant contrast between the shaved and non-shaved areas with a difference in CPD of 490 mV, which equates to a reduction in the work function of pristine MoS₂ of 0.49 \pm 0.03 eV. This indicates that functionalisation of monolayer MoS₂ with PDI reduces the surface potential and thus the work function, possibly due to doping. This has been previously observed for the work function of chlorine-doped CVD-grown MoS₂, where the doping reduced the φ_{MoS_2} by 0.4 eV as measured by KPFM.²¹²

The capability of nanoshaving as a lithographic technique is further confirmed by the near-complete absence of PDI's characteristic peaks in the Raman maps (Fig. 4.4(d)) and Raman and PL spectra (Fig. 4.4(f) and (g)). In the non-shaved areas, the characteristic signature for PDI molecules is clearly visible when resonantly excited at 532 nm (Fig. 4.4(f) and (g)). It should be noted the PDI signal is mostly absent on the SiO₂ substrate (except in areas of pileup), indicating that changes in step height above the substrate can be attributed to the SAM. The peaks at 1300 and 1380 cm⁻¹ are the in-plane ring 'breathing' modes, with the ring deformation occurring at 1457 cm⁻¹. The 1585 cm⁻¹ doublet is C-C stretching mode and the additional peaks visible in Fig. 4.4(g) between 1.95 and 2.05 eV (2500 - 3000 cm⁻¹) are overtones and combinations related to the first order vibrational modes.^{102,199,213,214} The Raman map of the MoS₂ A_{1g} mode (~ 404 cm⁻¹) in Fig. 4.4(e) shows no damage to the TMD underlayer from nanoshaving. This is further confirmed in Fig. A.2 where there is no change (i.e. no broadening) in the FWHM of both the E_g and A_{1g} peaks. In Fig. 4.4(g), the pristine MoS₂ PL at 1.85 eV is significantly quenched upon addition of the PDI. This quenching of PL while in a heterostructure has been observed by Obaidulla *et al.* for a similar

configuration, N,N'-diphenyl-3,4,9,10-perylendicarboximid (PTCDI-Ph) on MoS_2 .²⁰⁹ It was suggested that this effect was due to 'trap-like' states occurring at the hetero-interface, indicating separation of photogenerated electron-hole pairs, as well as less ordered morphology than PTCDA/MoS₂. Nevertheless, the MoS₂ PL intensity can be somewhat recovered upon nanoshaving the PDI (Fig. 4.4(g)). This reversibility reaffirms nanoshaving as a suitable candidate for lithography of these systems.

4.3.3 Further PL Analysis of Nanoshaved $MoS_2 & PDI$ on MoS_2

To evaluate the effect of the adsorbed PDI layer on the PL spectrum of the combined system, the PL spectra were further analysed and split up into their components using Pseudo-Voigt fitting in Origin software. The Pseudo-Voigt function is an approximation for the Voigt function, which is a linear combination of Gaussian and Lorentzian functions and accounts for the thermal broadening of the natural Lorentzian line shape.

Fig. 4.5(a) shows the fitted PL spectra (originally from Fig. 4.3(h)) of pristine MoS₂ (bottom) and nanoshaved MoS₂ (top). In Fig. 4.5(a), we can see that the A exciton (A^0) at ~ 1.83 eV decreases with intensity upon nanoshaving but the B exciton at ~ 1.95 eV remains the same. Neither appear to shift in energy but the ratio between the A^0 and B increases with nanoshaving from 0.04 to 0.05. There is a third component at $\sim 1.78 - 1.82$ eV, which is the formation of a negatively charged exciton or trion, due to interplay between the exciton and a charge carrier.¹⁷¹ Previous reports suggest the trion (A^-) recombination tends to dominate in mechanically exfoliated monolayer MoS₂ due to unintentional heavy n-type doping.^{171,215,216} However, in these spectra, the neutral exciton A^0 dominates the signal peak ~ 1.83 eV, which strongly suggests that the excitons can recombine without forming trions due to less doping/excess carriers in the CVD-grown monolayer. With nanoshaving of as-grown CVD MoS₂, the A^- signal appears to decrease very slightly in intensity. This combined with the slight quenching of the direct A⁰ recombination peak could be due to local compressive strain caused by nanoshaving, ²¹⁷ and removal of adsorbates. ²¹⁸



Figure 4.5: Fitted PL spectra of nanoshaved (top) and non-shaved (bottom) areas of **(a)** pristine MoS₂ from Fig. 4.3(h) and **(b)** PDI on MoS₂ from Fig. 4.4(g).

Fig. 4.5(a) shows the fitted PL spectra (originally from Fig. 4.4(g)) of PDI on MoS₂ (bottom) and nanoshaved MoS₂ (top). As previously mentioned, upon functionalisation of MoS₂ with PDI, several Raman modes of the perylene molecules are distinctly visible at $\sim 1300 - 1585$ cm⁻¹ ($\sim 2.13 - 2.18$ eV), of which overtones are seen at lower energies (2500 - 3000 cm⁻¹/1.96 - 2.02 eV) when excited at 532 nm. It should be noted, these distinct features are only visible on the surface of the TMD monolayer. PDI on the surface of the SiO₂ substrate does not demonstrate any Raman

signal but produces a broad PL signal at ~ 2.12 eV (see Fig.. A.3). The enhanced PDI Raman modes arise from orbital overlap with the TMD band structure, resulting in band alignment and energy transfer under resonant conditions.^{209,219} Furthermore, the enhancement of the B exciton (~ 1.97 eV) upon addition of PDI arises from the coupling of PDI's molecular excitons to the TMD excitonic states.²²⁰

If we examine the overall A exciton peak, we can see it is significantly quenched and also broadened (from 0.064 to 0.071 eV FWHM) by the PDI functionalisation. By including the PL from PDI on the SiO₂ (see Fig. A.3) in the peak model applied for pristine MoS_2 layers, the quenching and broadening can be attributed to a reduction in the A⁰ exciton with an increase in the A⁻ intensity. This reduction in the neutral exciton can be attributed to the introduction of trap-like states by the PDI adlayer and defects in the ordering on the MoS_2 surface.²⁰⁹

Though some evidence of PDI remnants on the MoS₂ monolayer upon nanoshaving is shown through the presence of the polyaromatic Raman modes ($\sim 2.13 - 2.18 \text{ eV}$), the PL emission of the nanoshaved area bears a resemblance to the spectrum of pristine MoS₂. Firstly, the PL spectrum shows an almost complete reduction in the PDI Raman signal, as well as the coupled PDI signal at $\sim 2.12 \text{ eV}$. Secondly, the enhancement of the B exciton by PDI coupling is also somewhat reversed, with a reduction in relative intensity from 2.37 to 1.57 post-shaving. Thirdly, the quenching of the overall A exciton by PDI is also reversed with a significant increase in intensity of A⁰. In contrast, the trion A⁻ is effectively quenched to beyond that of the unshaved pristine MoS₂. A potential explanation of this effect could be the passivation of MoS₂ defects by non-covalent functionalisation with PDI before subsequent removal, as has been seen to occur with PTCDA as a SAM layer on 2D TMD surfaces.²²⁰ However, to better observe the PL components, it is desirable to carry out measurements at lower temperatures (e.g. 77 K) to reduce non-radiative recombination effects on the spectrum.



4.3.4 PATTERNING PDI ON MOS₂ VIA NANOSHAVING

Figure 4.6: Nanoshaving grating pattern into PDI on MoS_2 – (a) topography and (b) CPD of PDI-functionalised CVD MoS_2 flake with nanoshaved grating pattern. (c) Topography and (d) CPD zooms of grating area showing uniformity. (e) Plot of height profile and corresponding potential profile (areas indicated in (c) and (d)). (f) average Raman spectra of shaved and non-shaved areas. Corresponding Raman peak intensity maps of (g) $MoS_2 A_{1g}$ (400 cm⁻¹) and (h) PDI (~1300 cm⁻¹) modes.

To demonstrate the applicability of this technique for lithography in these systems, a grating pattern was nanoshaved into the PDI/MoS₂ (see Fig. 4.6(a)). This resulted in a remarkably uniform grating (Fig. 4.6(b)), consisting of a ~ 360 nm pitch (250 nm trough, 110 nm peak). The average PDI pitch step height (Fig. 4.6(c)) is much higher (~ 8 nm above MoS₂) compared to unpatterned PDI on MoS₂, most likely due to shaving debris pile-up. From KPFM in Fig. 4.6(d), the surface potential corresponds precisely to the topography, which is evident when overlayed in Fig. 4.6(e). It shows a relative surface potential or work function difference between the PDI lines and MoS₂ of 600 mV or 0.6 eV, respectively, reaffirming the localised reduction in φ_{MoS_2} due to PDI functionalisation. This selective removal is also visible in the Raman spectra, with Fig. 4.6(f) showing a clear reduction in intensity of the PDI vibrational modes (~ 1300 cm⁻¹, ~ 1380 cm⁻¹ and ~ 1455 cm⁻¹). Nevertheless the PDI signal remains to some degree as the pitch is comparable with the laser spot size (~ 300

nm). Again, Fig. 4.6(g) shows the underlying MoS_2 to be undamaged and the parallel lines of PDI SAMs still adsorbed to the TMD surface (Fig. 4.6(h)). These results indicate that nanoshaving is a viable lithography technique introducing minimal contaminants, i.e. no resist necessary and without collateral damage to the SAM. In addition, nanoshaving could be suitable for the fabrication of other ordered optical/photonic structures, e.g. by rotating the sample 90°, a shaving pass of the same spacing parameters would produce a dot array.



4.3.5 Co-functionalisation of MoS_2

Figure 4.7: Co-functionalised MoS_2 – (a) Topography of PDI functionalised MoS_2 . (b) Topography and (c) CPD of nanoshaved PDI/MoS₂ area indicated by white dashed box in (a). (d) Topography and (e) CPD of area in (b) after deposition of PTCDA. (f) Topography and (g) corresponding CPD close-up of PTCDA/PDI boundary on MoS_2 .

A PDI/MoS₂ sample was further functionalised after nanoshaving with PTCDA to demonstrate chemically distinct non-covalent co-functionalisation of MoS₂. For the sake of clairty, the KPFM will be compared and examined first, followed by comparison of the Raman/PL pre- and post-deposition of PTCDA.

Fig. 4.7(a) shows an AFM topography image of MoS₂ flakes functionalised with PDI. In Fig. 4.7(b),

KPFM of the area outlined in Fig. 4.7(a) is shown post-nanoshaving with the corresponding CPD image, Fig. 4.7(c). The step height from PDI to the nanoshaved area on MoS₂ is ~ 1.3 nm, again consistent with previous SAM measurements.¹⁰² Fig. 4.7(c) shows significant contrast between the shaved and non-shaved areas with a difference in CPD of 390 ± 150 mV, which equates to an increase of 0.39 ± 0.15 eV in the φ of the PDI/MoS₂ heterostructure upon removal of PDI.

Fig. 4.7(d) shows the topography of the same flake after dropcasting PTCDA and Fig. 4.7(e) shows the corresponding CPD image. Both images indicate that a PTCDA SAM has formed on the nanoshaved area of MoS_2 . The large particulates observed were later determined to be excess PTCDA and were removed following further rinsing in DMF. Fig. 4.7(f) and (g) show a close-up KPFM scan of the PDI/PTCDA boundary on the monolayer flake. The PTCDA appears to form a closer-packed and more homogenous SAM than the PDI. This is most likely due to cleaning of the MoS₂ surface with nanoshaving before deposition of PTCDA, whereas PDI was deposited on an as-grown surface with surface contaminants. This further demonstrates the capability of nanoshaving as a local surface cleaning technique. Additionally, sterics on the sidechains, like in PDI, will effect packing. Fig. 4.7(g) shows the CPD contrast between the two heterostructures, with $arphi_{\it PDI/MoS_2}~=~5.87\pm0.11~{
m eV}$ and $arphi_{\it PTCDA/MoS_2}~=~5.96\pm0.10~{
m eV}.$ The slightly increased surface potential and thus work function of the PTCDA/MoS2 system can be attributed to the increased molecular order and packing density in the PTCDA SAM, which has previously been observed in KPFM of octadecyltrichlorosilane [OTS, CH3(CH2)17SiCl3] SAMs on oxidized Si(100) and polycrystalline silicon surfaces.²²¹


Figure 4.8: Co-functionalised $MoS_2 - (a)$ Raman peak intensity map of PDI ($\sim 1300 \text{ cm}^{-1}$) mode. (b) PL peak intensity map of MoS_2 A exciton ($\sim 1.85 \text{ eV}$). (c) average Raman spectra of shaved and non-shaved areas, showing characteristic PDI peaks. (d) average PL spectra of shaved and non-shaved areas. (e) Raman peak intensity map of the combined degenerate breathing, ring distortion and Kekulé mode²¹⁴ of PDI and PTCDA (1300 cm^{-1}) mode. (f) PL peak intensity map of MoS_2 A exciton ($\sim 1.85 \text{ eV}$). (g) Raman peak intensity map of PTCDA ($\sim 1340 \text{ cm}^{-1}$) mode. (h) Average Raman spectra of PDI and PTCDA functionalised areas, showing characteristic perylene peaks vs. the MoS_2 Raman peaks.

The non-covalent co-functionalisation of MoS_2 was further examined using the complementary optical techniques of Raman and PL. Like previous PDI shaving, the almost complete absence of PDI's characteristic peaks can be seen in the Raman map (Fig. 4.8(a)) and spectra (Fig. 4.8(c) and (d)), indicating that the SAM is fully removed by way of nanoshaving. In Fig. 4.8(b), the pristine MoS_2 PL at 1.85 eV is significantly quenched upon addition of the PDI. The PL spectra in Fig. 4.8(d) show the same features as described in Section 4.3.3 with the shaved and non-shaved areas. These reaffirm the KPFM measurements in demonstrating the almost complete removal of the PDI adlayer from the monolayer MoS_2 using nanoshaving.

As has been previously observed for PDI, the enhanced PTCDA Raman modes arises from orbital overlap with the TMD band structure, resulting in band alignment and energy transfer under resonant conditions.^{209,219} Fig. 4.8(e) shows the Raman peak intensity map of the combined

degenerate breathing, ring distortion and Kekulé mode of the perylene core²¹⁴ at ~ 1300 cm⁻¹, which is shared by both PDI and PTCDA. Fig. 4.8(f) also shows quenching the MoS₂ A exciton in the shaved area, similar to the quenching by PDI. In Fig. 4.8(g) a vibrational mode unique to PTCDA (~ 1340 cm⁻¹) is mapped, along with the Raman spectra of the PDI and PTCDA functionalised areas in Fig. 4.8(h). All of these indicate the formation of a PTCDA SAM solely on the nanoshaved region of MoS₂, thus demonstrating co-functionalisation of TMDs, while the PDI adlayer helps to passivate the other areas towards PTCDA assembly.

It should be noted that the PTCDA Raman signal is many orders of magnitude more intense than the MoS₂ E_{2g}^1 and A_{1g} modes (~ 384cm⁻¹ and ~ 405 cm⁻¹), and the PDI modes (1300, 1380, 1457 cm⁻¹ and the doublet at 1585 cm⁻¹). This leads to any PTCDA contamination on the PDI SAM dominating the signal, hence the large overlap in spectra in Fig. 4.8(h) of the PDI and PTCDA SAMs.

4.3.6 Change in Work Functions with Nanoshaving & Functionalisation

CPD measurements from all flakes were compiled for statistical purposes to produce Fig. 4.9(a) and (b), detailing the work function ranges for each of the parameters and the corresponding change in work functions from pristine CVD-grown MoS₂. Nanoshaving pristine MoS₂ does not appear to significantly alter the work function while PDI seems to lower the pristine work function by ~ 0.45 eV. In contrast, the addition of PTCDA on the nanoshaved MoS₂ appears to lower φ_{MoS_2} by only ~ 0.23 eV. The mean experimental φ_{PTCDA/MoS_2} value of 5.62 eV, along with the mean φ_{MoS_2} value of ~ 5.85 eV, are comparable to the calculated work functions of 5.946 eV for pristine MoS₂ and 5.725 eV for PTCDA/MoS₂ heterostructure by Habib *et al.* ²⁰³ This theoretical $\Delta \varphi$ of 0.221 eV between pristine MoS₂ and the heterostructure is in strong agreement with the experimental $\Delta \varphi$ of 0.23 eV in this work. Habib *et al.* attributes this change in work function after introducing an



Figure 4.9: (a) Work function range as measured by KPFM. (b) change in work function from pristine MoS_2 with nanoshaving and non-covalent functionalisation.

organic over-layer on the inorganic surface to a chemical effect, the change in dipole moment at the interface due to charge rearrangement, adsorption induced bandbending effect at the interface and overlayer induced substrate relaxation.^{203,222–224} The spread in CPD values between flakes and measurements in this work can in part be attributed to electrical isolation. However, as any potential offset should be the same on any single flake, a comparison can still be made to determine the difference in potential between shaved areas and non-shaved areas. This is providing that any changes in the state of the tip are accounted for through calibration, as was the case here.

A spread in work functions determined by KPFM is not unexpected as it is known that such values are highly sensitive to numerous factors including surface contamination, sample quality and grounding and experimental conditions (ambient in this case of this work). In literature, this leads to a relatively large spread in reported work function values of MoS₂ calculated from CPDs. Choi *et al.* investigated the layer-dependent work function of MoS₂ by way of surface potential measurements in KPFM.²²⁵ The work functions of mechanically exfoliated (ME) 1, 2 to 6 layer MoS₂ were 5.15, 5.25 - 5.39 eV and increased with increasing layer number. After annealing, the flake work functions decreased by 0.1 - 0.2 eV due to the reduction in the amount of surface adsorbates. Tamulewicz *et al.* examined the influence of flake-substrate effects on the work functions of mono to few layer ME MoS₂, with values ranging from 4.84 eV for monolayer, 4.89 eV for bilayer to 4.99 eV for 5 layer.²²⁶ In contrast, Habib *et al.* calculated the work function for pristine ML MoS₂ to be 5.946 eV,²⁰³ which is very comparable to the mean experimental φ_{MoS_2} value in this study of ~ 5.85 eV. While still within range of the former two studies, 5.85 eV is relatively high in comparison with the aforementioned values. One explanation might be the fabrication methods: mechanically exfoliated flakes (top-down) vs. CVD-grown monolayers (bottom-up), though a more in-depth comparison study is needed.



4.3.7 NANOSHAVING PDI ON WS₂

Figure 4.10: Nanoshaving PDI on WS2 – (a) topography and (b) CPD of nanoshaved PDI functionalised WS₂. Raman peak intensity maps of (c) PDI (1300 cm⁻¹) and (d) WS₂ 2LA(M) + E_{2g}^1 mode (352 cm⁻¹). (e) PL peak intensity map of WS₂ A exciton (1.95 eV) with (f) corresponding average spectra of functionalised vs. shaved areas.

To demonstrate that these methods could be applied to different TMD systems and not just limited to MoS₂, PDI was deposited on WS₂ (in the same fashion as MoS₂) and an area of flake was

subsequently nanoshaved (Fig. 4.10). The average load applied was 1.08 µN, slightly less that the average load of 1.25 µN used on PDI/MoS₂. As previously mentioned, the step height of pristine CVD-grown WS₂ monolayer was determined to be 0.75 nm by AFM. Upon addition of PDI, the step height of the heterostructure was measured to be ~ 1.9 nm (Fig. 4.10(a)), similar to PDI/MoS₂ and comparable to previous observations.¹⁹⁷ The contrast in CPD between nanoshaved and non-shaved areas is clearly visible in Fig. 4.10(b), with a difference of ~ 400 mV, equating to a reduction in φ_{WS_2} of 0.4 eV, from an average $\varphi_{WS_2} = 6.13$ eV to $\varphi_{PDI/WS_2} = 5.73$ eV. Like MoS₂, the mean pristine WS₂ work function is much higher (~ 1 eV) than than previously reported values.^{227–229} However, our observation of $\varphi_{WS_2} > \varphi_{MaS_2}$ is in agreement with literature values.^{228,229}

The complete removal of PDI was further confirmed by Raman analysis in Fig. 4.10(c). Pristine WS₂ exhibits a characteristic peak at 352 cm⁻¹, which, when excited at 532 nm, is a combination of several different modes (see Fig. A.4). Of these, the most intense are the in-plane vibrational mode (E_{2g}^1) and the resonantly excited 2LA(M) phonon mode.^{145,151} This intense peak can be seen in Fig. 4.10(d) indicating negligible damage from nanoshaving PDI to the underlying WS₂ monolayer. Pristine monolayer WS₂ also exhibits a strong RT PL response at ~ 1.95 eV, which can be seen in Fig. 4.10(e). As seen for MoS₂, the significant quenching of WS₂ PL by functionalisation with PDI is observed in Fig. 4.10(f). This quenching effect of PDI is even more substantial in WS₂ than in MoS₂, which warrants further investigation.

4.4 CONCLUSIONS

Non-covalent functionalisation of monolayer TMDs, MoS₂ and WS₂, on Si/SiO₂ with organic SAMs of PDI was explored. These organic-inorganic heterostructures were manipulated using the

novel approach of nanoshaving and the reversibility of such functionalisation was investigated using the complementary non-destructive techniques of KPFM and Raman spectroscopy. Nanoshaving pristine monolayer TMD flakes had negligible effect on the topography but KPFM showed contrast in CPD between the shaved and non-shaved areas, most likely due to clearing of surface contaminants. It was shown that PDI selectively forms SAMs on TMD surfaces rather than Si/SiO₂. A grating pattern with feature sizes < 250 nm was successfully shaved into the PDI/MoS₂ heterostructure, demonstrating the high-resolution capabilities of nanoshaving as a lithographic technique. In addition, non-covalent co-functionalisation of MoS₂ was demonstrated using PTCDA SAMs on previously nanoshaved areas. We need to go back to the discovery, to posing a question, to having a hypothesis and having kids know that they can discover the answers and can peel away a layer.

Shirley Ann Jackson, physicist & first black woman to

earn a Ph.D. from MIT

5

Nanomachining PtSe₂

5.1 INTRODUCTION

THE RELENTLESS PURSUIT OF LAYERED transition metal dichalcogenide (TMD) materials derives from their unique properties, from semimetals to semiconductors, for use in 2D-based electronics.^{64,67,230,231} Recently, considerable interest has grown in group X TMDs, referred to as noble-metal-based TMDs (NTMDs).^{77,232–234} Their properties differ from the more widely studied group VI TMDs, such as dichalcogenides of Mo and W, due to the fact that, in NTMDs, all d-orbitals are filled, resulting in a d²sp³ hybridisation.²³⁵ Hence, NTMDs form a thermodynamically stable octahedral 1T phase (Fig. 5.1), unlike more commonly studied TMDs, which tend to form a stable trigonal prismatic 2H-phase.^{62,81} The electron-rich, non-bonding d-orbitals result in stronger coupling between the two p_z bands of two adjacent intermolecular chalcogen atoms, leading to the extraordinarily strong interlayer interactions with nearly isotropic in-plane and out-of-plane mechanical interlayer coupling.⁸¹ In the case of PtSe₂, this makes it challenging to isolate monolayer PtSe₂ micromechanically from bulk crystals.²³⁶ In addition, the interlayer breathing force constant, which describes the interlayer coupling, is calculated to be 2.5 times larger than that of graphene.¹⁴⁶

However, the isolation of mono- and bilayer PtSe₂ is of utmost interest for 2D-based electronics as it shows layer-controllable semimetal (indirect) semiconductor to transition.^{72,73,231,238} The monolayer (ML) has been reported as having an indirect band gap of 1.2 - 1.6 eV with the bilayer (BL), also indirect, of ~ 0.8 eV.^{75–78} This quantum confinement (number of layers) enables modulation of its electronic structure resulting in a tuneable bandgap.^{60,64,231,238} Moreover, PtSe₂'s band structure can be modified by external influences such as electric fields,²³⁵ doping,²³⁹ defect engineering²⁴⁰ and mechanical forces^{236,241,242}.



Figure 5.1: Summary of applications, properties and synthesis of $PtSe_2$.²³⁷ Below, three-dimensional schematic representation of a $PtSe_2$ (1T) monolayer structure, with selenium (Se) atoms in yellow and platinum (Pt) atoms in grey.

PtSe₂ has attracted much interest as a potential candidate for many applications, including field-effect transistors (FETs),^{79,96} optoelectronics,^{243,244} catalysis^{245,246} and chemical and piezoresistive sensors.^{80,247,248} Zhao *et al.* reported that few-layer PtSe₂ FETs exhibit high electron mobility at room temperature (RT) ($\sim 210 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$) on SiO₂/Si substrate.⁷⁹ Nevertheless calculations predict significantly higher mobilities of $\sim 4000 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ for PtSe₂ at RT.⁷⁸ In addition, PtSe₂ remains stable in air over elongated periods of time.⁷⁹

Despite such progress in the research of NTMDs, challenges still remain, particularly in the understanding of the synthesis and properties of NTMD thin films. Precise control over the material dimensions, while simultaneously maintaining high material quality, is paramount to the study and understanding of thickness-dependent properties. In addition, given its application potential, investigating large-scale production methods in line with standard semiconductor back-end-of line (BEOL) processing is key to meeting future industrial integration needs.^{96,249} Although high-quality few layer PtSe₂ flakes can be micromechanically exfoliated from bulk crystals,^{79,250–252} the lack of scalability of the approach makes them only suitable for fundamental studies. Besides the top-down methods of mechanical exfoliation (ME) and liquid-phase exfoliation (LPE),²³⁵ thin films of PtSe₂ have been fabricated *via* a range of bottom-up techniques such as thermally assisted conversion (TAC),^{72,76,96,231,247,253–255} chemical vapour deposition (CVD)^{233,256–258} and molecular beam epitaxy (MBE).²⁵⁹

Controlled physical modification of NTMDs is key to investigating their fundamental properties for use in numerous nanotechnology-based applications. As discussed in Chapter 4, scanning probe microscopy (SPM) techniques have been widely employed in the fabrication of nanostructures particularly on 2D materials, which AFM being the technique in this work. While nanoshaving selectively removes adsorbed layers (adlayers), ⁵¹ nanomachining is when a sufficiently stiff SPM tip is used to ablate a surface mechanically with a required normal force, removing materials locally from

the sample. ^{50,52} This technique has been exploited in a wide variety of studies, such as fabricating complex arrays of 3D nanodots on polycarbonate samples, ²⁶⁰ nanochannels of varying depth in silicon, ²⁶¹ ferroelectric structures ⁵² and investigating the depth distribution of nitrogen-vacancy (NV) centers formed in an irradiated diamond surface. ²⁶² AFM nanomachining also enables data to be collected *in situ* during the nanoshaving/nanomachining process, giving insight into the mechanical properties of the material.

Again, as previously discussed in Chapter 4, the mechanical nature of the nanomachining process is the root of its limitations. The limiting factor in creating reproducible patterns is the stability of the tip, which is prone to deformation and contamination from debris of the removed material,⁴⁷ and ridge formation or the accumulation of materials around the inscribed features.⁵⁴

In terms of NTMDs, ML PtSe₂ is relatively robust in nature as its structure changes negligibly under biaxial tensile deformation,⁷⁵ making it a suitable candidate for nanomachining, particularly with its layer-dependent electronic structure. While ML and BL 1T-PtSe₂ are indirect SCs, bulk and few layer 1T-PtSe₂ exhibit semi-metallic in behaviour, making it a very interesting candidate for the possibility of 'self-contacting' FETs, i.e. where the semiconducting channel and contacts are made of the same material, simplifying the devices.

A major obstacle in the development and integration of practical 2D electronic and optoelectronic devices which must be addressed is that of reliable, low-resistance electrical contacts to 2D materials.¹³⁰ It has been shown that large contact resistances can occur between TMD channels and bulk metal contacts, significantly reducing the efficiency of current flow.¹³¹ Regardless of the electrode material used, it requires the transmission of electrons through the interface between two different crystal lattices, which always causes a contact resistance due to reflections, Schottky barriers and scattering at the interface.¹³² Low contact resistance in 2D SC devices is crucial for achieving high 'on' current, large photoresponse¹³³ and high-frequency operation.¹³⁴ Nevertheless, the chemical interaction at the metal-SC interface governs everything for 2D materials. Pristine surfaces

of 2D materials (i.e. no dangling bonds) hinder formation of any interface bonds with a metal, thus increasing contact resistance.¹³⁵

As it is difficult for metals to covalently bond with pristine 2D material surfaces, a van der Waals (vdW) gap forms at the interface, acting as an additional tunneling barrier for the charge carriers, in addition to any innate Schottky barrier.^{130,136} This extra barrier reduces the charge injection from metals, which leads to higher contact resistance. By reducing this vdW tunnel barrier as well as optimising edge contacts with all layers of the 2D material to produce a more seamless contact design, the contacts can be significantly improved with reduced resistance.



Figure 5.2: Schematic representation of a PdS₂ diode, consisting of a BL/ML/BL junction, with sulfur (S) atoms in yellow and palladium (Pd) atoms in brown. ¹³² l_c and l_o refer to the channel and overlap lengths.

An alternative to conventional metal contacts has been proposed *via* phase-engineering whereby a single-layer 2H MoS₂ transistor was contacted with metallic 1T MoS₂.²⁶³ However, 1T MoS₂ is metastable and reconverts to the 2H phase at room temperature (RT), thus negating the electrodes and device. Similarly, a single-material, ultrathin device based on 1T PdS₂ was proposed theoretically by Ghorbani-Asl *et al.*,¹³² which consisted of a logic junction based on ML (band gap of ~ 1.1 eV) and BL (semimetallic) PdS₂ (Fig. 5.2). It was shown that for channel lengths $l_c \geq 2.45$ nm, the leakage current becomes negligible; any shorter and the junctions show tunnel diode characteristics. The concept is very attractive for energy efficient electronics as the whole device is built on a single flake and crystal plane, resulting in low contact resistance and sustainability (only one material used).¹³² As PtSe₂ has similar properties to PdS₂, a device based on a PtSe₂ channel and contacts

constructed in this manner should be possible with appropriate fabrication methods.

5.2 Experimental Methods

5.2.1 THERMALLY ASSISTED CONVERSION (TAC) DEVICES

For thermally assisted conversion (TAC) of platinum to platinum diselenide, the platinum metal thin films were deposited by physical vapour deposition (PVD) onto Si/SiO₂ substrates with a 300 nm oxide layer. Three methods of PVD were used to deposit a controlled thickness of Pt; sputtering, electron beam evaporation and molecular beam epitaxy (MBE). The sputter coater used was a Gatan Precision Etching and Coating System (PECS) with a MaTeck Pt target. For e-beam evaporation of Pt, a Temescal FC-2000 Evaporation System was used. Both sputtering and e-beam evaporation use a quartz crystal microbalance to monitor deposition rate and film thickness.

Growth of Pt thin films were carried out in a DCA M600 MBE system with a base pressure of 5×10^{-10} Torr on a variety of Al₂O₃ growth substrates; c-plane sapphire and amorphous alumina. The substrates were annealed under vacuum for ~ 2 hours prior to deposition. Pt was deposited at 600 °C. The thin films were subsequently selenised *via* the TAC method.

Once appropriate thicknesses^{*} of Pt were deposited, the thermally assisted conversion (TAC) process, as described previously by O'Brien *et al.*⁷⁶, was utilised to transform the metal into the TMD, PtSe₂.

Lithographic techniques were used to create channel devices for electrical characterisation (see Fig. 5.3). Two approaches were used: photolithography (UVL) and electron beam lithography

^{*}From atomic force microscope (AFM) measurements of the thicknesses of Pt layers before and after selenisation, it has been reported that the initial Pt thickness expands approximately four times after post-selenisation.⁹⁶ Henceforth, TAC film thicknesses will be referred to by their starting Pt thickness unless explicitly stated otherwise.



Figure 5.3: Schematic of the fabrication of $PtSe_2$ thin film channels via TAC.⁹⁶

(EBL). The lithography tool used was an OAI mask aligner situated in a Class 100 cleanroom. This generally involved two steps in fabrication: first - markers and channels, and second - contacts (post-selenisation). Three steps were generally involved in fabricating the channel devices using EBL; markers, channels and then contacts post-selenisation. For both techniques of device fabrication, positive photo resist was used to allow precise metal deposition. Once, the metal was deposited, the remaining resist was removed in a lift-off, leaving clearly defined channels and subsequently contacts.

5.2.2 MECHANICALLY EXFOLIATED (ME) DEVICES

To prepare samples with ME flakes on the substrate, polydimethylsiloxane (PDMS) stamps (Gel-Pak-4) were used in transferring the thinned flakes from scotch tape to SiO₂/Si substrates using a transfer stage. Substrates were pre-patterned with Au coordinates for ease of EBL/UVL. Once the flakes had been transferred from the stamp to the substrate, the sample was annealed under Ar at 150 °C to ensure good adhesion for spin-coating purposes. Devices were fabricated using maskless UVL as described in Section 3.3. Ti/Au contacts (5/45 nm) were deposited on the ME flakes to create a two terminal device.

5.2.3 CHARACTERISATION & MANIPULATION

AFM topography and nanomachining were performed in ambient conditions using an Asylum Research Cypher and a Bruker Multimode 8. TAC films and ME flakes were nanomachined at different scan speeds and setpoints to optimise parameters using a Bruker Multimode 8. All nanomachining operations were carried out using wear-resistant nanomechanical diamond cone tips (NM-RC, k = 600 N/m, $f_0 = 600 \text{ kHz}$, supplied by Adama Innovations Ltd.) in contact mode. The deflection sensitivity of the cantilevers was determined by performing force-distance curves pre-shaving on a sapphire substrate. All KPFM and PF-TUNA measurements were carried out using SCM-PIT-V2 probes from Bruker which are electrically conductive and coated with PtIr. For KPFM, the tips were calibrated using a Au-Si-Al calibration sample (PFKPFM-SMPL, Bruker). To minimise possible topographical artefacts, PeakForce KPFM (PF-KPFM), a two-pass (lift) mode, was used, where topographical data was recorded in the first pass. Then the tip was lifted to a user-specified distance above the sample surface (~ 85 nm for SCM-PIT-V2), measuring work function in the second pass. Samples were also electrically connected directly to the chuck through silver paint (Agar Scientific) contact from the sample to its magnetic puck. The subsequent images were analysed and their topographical profiles extracted using the Gwyddion software.

Raman spectroscopy was carried out using a WITec Alpha 300R spectrometer with a 532 nm excitation laser, which was fitted with a Rayshield Coupler to detect Raman lines close to the Rayleigh line at 0 cm⁻¹. Spectra were recorded using a laser power of $< 300 \,\mu\text{W}$ in order to minimise sample heating. A 100x objective with a numerical aperture (NA) of 0.95 and a spectral grating with 1800 lines/mm were used for all Raman spectra. Maps were generated by taking 4 spectra per μ m in both x and y directions over large areas. The resulting data was analysed and the spectra were subsequently graphed using a combination of WITec Project FIVE and Origin Pro software.

Electrical characterisation was performed in ambient conditions using a Karl Suss probe station connected to a Keithley 2612A source meter. The voltage was applied through needle-probes, contacting Au electrodes deposited on the TMD films/flakes. The measured current was collected and plotted automatically using Labview and Origin Pro software to give current-voltage (I-V) curves.

5.3 RESULTS AND ANALYSIS

5.3.1 NANOMACHINING TAC PTSE₂

PtSe₂ thin films were grown using the TAC process described in Section 3.2.3 whereby a Pt film on SiO_2/Si was reacted with Se vapour at elevated temperatures in a low-pressure environment. As previously mentioned in Section 5.2.1, TAC film thicknesses will be referred to by their starting Pt thickness. The post-selenisation expansion factor is approximately four times the deposited thickness of Pt.⁹⁶ These films were characterised by AFM (see Fig. 5.4), which revealed their polycrystalline nature. The roughness (root mean square, RMS) of 3 nm converted PtSe₂ film was determined to 3.4 nm.[†]

Blanket films were imaged using tapping mode (Fig. 5.4(a)) to examine the region prior to nanomachining. Then a $1 \times 1 \mu m$ area was machined while in contact mode at a predetermined setpoint of 280 mV and scanspeed of $3.9 \mu m/s$. From Fig. 5.4(b), it is clear that the setpoint was too high and thus the PtSe₂ film was completely removed down to the SiO₂, resulting in a large pile-up of material (Fig. 5.4(d)). It would seem the oxide layer was also partially machined as the post-selenisation thickness would be ~ 12 nm and the machining depth was ~ 15 nm (Fig. 5.4(c)).

[†]These images were produced using a broad angle diamond nanomachining tip, hence the 'haloing' effect on the topographical AFM images.

Another test was conducted whereby the setpoint was gradually increased by 100 mV per 100 nm of nanomachining (scanspeed remained constant at 3.9 μ m/s) shown in Fig. 5.4(e). The resulting line profile (Fig. 5.4(f)) indicates a clear correlation between increasing setpoint and nanomachining depth, with ~ 1 nm of material removed per 100 mV increase.



Figure 5.4: (a) Topographical image of 3 nm PtSe₂ TAC film. (b) TAC film post-nanomachining (5 μ N load, 20 μ m/s tip velocity. Arrow indicates rastering direction during machining. (c) Line height profile of machined region indicating a machining depth of \sim 15 nm. (d) 3D rendered image of (b) showing pile-up. (e) Post-nanomachining with incrementally increasing setpoint. Line indicates height profile. (f) Averaged height profile of machined region in (e) indicating increasing machining depth with increasing setpoint.

NANOMACHINING EBL-DEFINED TAC CHANNEL DEVICES

SiO₂/Si substrates were patterned using EBL to create an array of channel devices with channel widths of 1000, 800, 500, 200 nm. 3 nm Pt was then sputtered and subsequently selenised *via* TAC. Topographical AFM line profiles determined the post-selenisation thickness to be ~ 12 nm. Contacts were patterned, again using EBL, whereby 5:35 nm Ti:Au was then evaporated onto the substrate. Channels were nanomachined at a variety of scanspeeds (1 - 40 μ m/s) and setpoints

(10 - 500 mV) to optimise parameters.



Figure 5.5: (a) Topographical image of 3 nm PtSe₂ EBL TAC channel (width 1000 nm) between TiAu contacts. (b) TAC channel post-nanomachining (3.8 μ N load, 20 μ m/s tip velocity). (c) Line height profile of machined region between contacts indicating a contact height to machining depth of \sim 40 nm. (d) Line height profile normal to the channel showing \sim 5 nm machining depth. (e) SEM of channel post-nanomachining. (f) Zoom of machined region in (f) showing complete removal of channel in some areas.

The first channel (Fig. 5.5(a)) was nanomachined normal to the channel length (Fig. 5.5(b)), leading to pile-up in between the contacts. Here the setpoint was deemed too high at 3.8 N, resulting in almost complete removal of the film. The topographical line profile in Fig. 5.5(c) (indicated by the white line normal to the TiAu contacts in Fig. 5.5(b)) shows the height of the contacts relative to the machined surface to be ~ 40 nm, further confirming the removal of the 12 nm TAC film. In Fig. 5.5(d), the topographical profile (indicated by the white line parallel to the TiAu contacts in Fig. 5.5(b)) the machining depth is ~ 5 nm, with a pile-up being ~ 70 nm in height, both relative to the SiO₂ surface. Fig. 5.5(e) and (f) are both SEM images of the machined channel, showing the removal of the PtSe₂ TAC channel.

The inter-contact pile-up in Fig. 5.5(b) could have potentially lead to shorting the device, thus, for

the subsequent channel (Fig. 5.6(a)), the machining procedure was rotated by 90° (Fig. 5.6(b)). In addition, the setpoint was reduced to 2.6 N, resulting in a thinner smoother TAC film with a reduction in RMS roughness from 2.55 nm pre-machining to 1.16 nm.



Figure 5.6: (a) Topographical image of 3 nm PtSe₂ EBL TAC channel (width 800 nm) between TiAu contacts. (b) TAC channel post-nanomachining (2.6 μ N load, 20 μ m/s tip velocity). (c) Line height profile normal to the channel showing \sim 8.5 nm machined channel height. (d) Line height profile along channel indicating \sim 5 nm machining depth. (e) SEM of channel post-nanomachining. (f) Zoom of machined region in (f) showing machined channel.

Fig. 5.6(c) shows the line profile (indicated by the white line normal to the $PtSe_2$ channel in Fig. 5.6(b)) of the machined channel of ~ 7 nm, indicating a height reduction of ~ 5 nm. In Fig. 5.6(d), the topographical profile (indicated by the white line parallel to the $PtSe_2$ channel in Fig. 5.6(b)) shows an average pile-up height of ~ 60 nm and further confirms the machining depth of ~ 5 nm. The SEM images in Fig. 5.6(e) and (f) show how the majority of the $PtSe_2$ channel remained intact post-machining and appears smoother, in agreement with what was measured in AFM. However, upon closer examination of Fig. 5.6(f), it appears that some grains were removed completely from film, causing discontinuities. In addition, initial roughness of TAC surfaces led to difficulties in reproducible machining, even though machining appeared to reduce roughness.

Attempts were made to perform I-V measurements on the machined channel devices but it was found that no current was transmitted. This was mostly likely due to discontinuity in TAC films in general but also exacerbated post-machining due to surface roughness caused by the TAC process (see Fig. 5.6(f)).

5.3.2 NANOMACHINING PT PRE-SELENISATION

An alternative method trialled was to nanomachine pristine Pt before selenisation to achieve desired thickness as TAC significantly increases roughness, particularly with thicker Pt films. Using UV lithography (Section 3.3), 30 nm of Pt was sputtered onto SiO₂/Si substrate (Fig. 5.7(a)). 30 nm was chosen as this is the general thickness for metal contacts. The quality and control over the machining was significantly improved due to the relatively smooth Pt surface (RMS roughness ~ 1.3 nm). Subsequently, for a 5 µN load and 20 µm/s tip velocity, it was possible to control the machining depth for each pass within a nanometer (Fig. 5.7(b) and (c)). In addition, a Pt channel was nanomachined repeatedly with the same parameters (5 µN, 20 µm/s), allowing for iterative thinning for multiple passes, as can be observed in Fig. 5.7(a),(d)-(f).

By the third pass (Fig. 5.7(f)), bulk Pt appears to have been removed from the edges of the channel, i.e. channel width has significantly decreased. Averaged topographical profiles were extracted from each of the images (a),(d)-(f) for comparison. Fig. 5.8(a) and (b) shows the average Pt channel heights and widths respectively before and after each machining pass. The decrease in the Pt channel width further confirms the delamination of Pt during the nanomachining process. This is most likely due to Pt's poor adhesion to SiO_2 , ^{264,265} rather than the nanomachining alone. Therefore, 30 nm Pt on SiO_2 is too thick a starting point for nanomachining. In Fig. 5.8(c), the step heights (from the pristine to machined Pt in Fig. 5.8(a)) are used to determine the recess height and the amount of material removed per pass relative to the order of machining passes. The mean amount of Pt removed with each pass at



Figure 5.7: (a) Topographical image of 30 nm thick Pt channel (width 3 μ m) between Pt (also 30 nm) contacts. (b) Pt channel post-nanomachining. Dotted line indicates machined region. (c) Line height profile showing \sim 1 nm machining depth. (d) Pt channel (a) after first pass, (e) second pass, and (f) third pass, each at 5 μ N load, 20 μ m/s tip velocity.

a constant load of 5 μN is ~ 3.83 nm.

It has been found that converting more than a few nm of Pt leads to disruption of the films, negating the pre-selenisation machining. This is as a result of the change in preferred orientation of PtSe₂ films forming during selenisation. Lin *et al.* reported an exponential increase in edge-rich (i.e. vertically aligned) PtSe₂ as a function of increasing thickness, from ~ 20 nm upwards²⁶⁶ (~ 5 nm for starting Pt thickness). Similar morphologies of layers perpendicularly aligned to the growth substrate have been seen in MoS₂ and MoSe₂.⁹⁵ Upon selenisation of > 5 nm of Pt, Lin *et al.* suggest chemical conversion occurs much faster than diffusion of Se gas through the film, i.e. diffusion along the layers through the vdW gaps would be significantly faster than through the layers, resulting in the layers naturally orientating perpendicular to the film. Therefore the in-plane electronic transport would be significantly decreased in PtSe₂ films thicker than 20 nm due to an increased contribution from out-of-plane pathways, such as vertically aligned grains.



Figure 5.8: Topographical profiles of Pt channel (width $\sim 3 \,\mu m$) before and after multiple nanomachining passes (each at 5 μN load, 20 μm /s tip velocity) (a) parallel and (b) perpendicular to the Pt channel. (c) Plot showing pass number vs. recess height (i.e. pristine Pt to machined step height) and amount of material removed per pass.



Figure 5.9: Topographical AFM images of Pt films deposited *via* molecular beam epitaxy (MBE). (a) 15 nm Pt despoited on c-plane sapphire (0001 Al₂O₃). (b) Scale-adjusted in (a) to show terraces of growth substrate. (c) Magnified image of 15 nm Pt MBE growth. (d) 1 nm Pt on C-plane sapphire pre-selenisation. (e) 1 nm Pt on amorphous alumina (Al₂O₃) pre-selenisation. (f) and (g) are (d) and (e) after conversion to PtSe₂ *via* the TAC method. Scale bar for (d) - (g) is 200 nm.

As previously noted, since Pt adheres poorly to SiO₂, it is worth examining other substrates in addition to higher crystallinity in the starting Pt film. An investigation into the substrate-dependent deposition of Pt and subsequent selenisation was also carried out. To provide better control, molecular beam epitaxy (MBE) was employed for growth of the Pt film, with the target substrate heated at an elevated temperature to 600 °C during deposition. While the epitaxial quality improves at higher substrate temperatures (> 600 °C), above a critical temperature Volmer-Weber growth mode starts and causes a rough film morphology.²⁶⁷ In addition, a comparison between growth substrates was carried out, namely amorphous alumina (Al₂O₃) and c-plane (0001) sapphire (monocrystalline Al_2O_3). Fig. 5.9(a) shows the epitaxial growth of Pt (111) on (0001) Al_2O_3 . The height scale in (a) is adjusted in Fig. 5.9(b) to show the (0001) Al₂O₃ atomic steps. Fig. 5.9(c) shows a close-up of the the continuous oriented film indicating crystalline growth with preferred epitaxial relationship and smooth surfaces, including step heights in agreement with literature values for atomic steps of Pt (111).²⁶⁸ In Fig. 5.9(d) and (e), we can see that the MBE deposition of Pt on the monocrystalline substrate forms larger grains than on the amorphous growth substrate. For thinner films, there is island growth (Volmer-Weber) rather than the ideal Frank-van der Merwe (layer-by-layer) or even Stranski-Krastanov (layer-plus-islands) growth. From Fig. 5.9(f) and (g), we can see that the deposited Pt grain sizes in (d) and (e) increase further upon selenisation for all substrates. For c-plane sapphire (0001 Al₂O₃), the mean grain diameter increases from 11 nm pre-selenisation to 40 nm upon conversion to the TMD. The same can be said for amorphous alumina, however albeit slightly less, from 7 to 20 nm.

5.3.3 NANOMACHINING MECHANICALLY EXFOLIATED PTSe₂ Flake Devices

For comparison with the roughness of the TAC films, pristine mechanically-exfoliated $PtSe_2$ flakes were nanomachined as a parallel approach. Bulk $PtSe_2$ crystals were obtained from HQ Graphene. Using the method mentioned in Section 3.2.5, pristine flakes were exfoliated using Scotch tape and later transferred to SiO₂/Si substrates. Due to the unusually strong interlayer interaction between $PtSe_2$ layers, problems arose while trying to isolate thin flakes that were large in lateral area (> a few μ m). Since the initial yield of thin flakes was very low, some preliminary machining experiments to calibrate the nanomachining rate were carried out on thicker flakes.

The nanomachined ME flakes indicate that, if the roughness of TAC films is decreased, the quality and control of the nanomachining process is greatly improved. On ME flakes, the shallowest machining depth was ~ 0.8 nm, approximately the thickness of one ML, showing the capability of nanomachining to achieve layer-by-layer removal.

After refinements in the ME technique, the yield of larger, thinner flakes improved. AFM images of such flakes can be seen in Fig. 5.10(a), (c) and (d). From line profiles of Fig. 5.10(a), a ML thickness of ~ 0.6 nm was determined (Fig. 5.10(b)). This was again confirmed by a comparison of topographical AFM and Raman spectra of the corresponding low-frequency layer-dependent peaks of the same area in Fig. 5.10(d) and the spectra are in good agreement with reported values.⁷⁹

Characteristic PtSe₂ peaks can be seen in the Raman spectra shown in Fig. 5.10(e) and (f). In (f) there are the two prominent vibrational modes visible. When excited at 532 nm, the $\sim 180 \text{cm}^{-1}$ peak corresponds to the in-plane E_g mode, while the $\sim 205 \text{ cm}^{-1}$ peak is the out-of-plane A_{1g} mode. In addition, thickness-dependent shifts of the E_g mode and changes in the E_g/A_{1g} intensity ratio are observed. Both the E_g mode redshifts and the A_{1g} mode increases in relative intensity to the E_g with increasing thickness.²³⁵ The less intense feature at $\sim 230 \text{ cm}^{-1}$ is assigned to an overlap between the A_{2u} and E_u modes, which are longitudinal optical modes involving the out-of-plane and in-plane motions of Pt and Se atoms respectively.⁷⁶ This contribution is most evident with close-to-ML thickness and decreases in relative intensity with increasing flake thickness.

For 2D layered materials, lattice vibrations contain high-frequency intralayer vibrations (e.g. E_g , A_{1g}) and low-frequency interlayer vibrations.¹⁴⁶ The interlayer vibrations are located at low-frequencies that cannot be resolved by standard equipment. These can be divided into the out-of-plane



Figure 5.10: Topographical AFM images of ME PtSe₂ flakes. (a) Bulk region of ME PtSe₂ flakes showing multiple step edges. White dashed line indicates ML step edge. (b) Topographical line profiles of (a) indicating ML step edge and thickness. (c) ME flake with 3L region. White lines indicate height profiles. (d) AFM image of ME flake with corresponding low-frequency Raman peak intensity map overlay indicating layer number. (e) Low-frequency Raman spectra from different regions in (c) and (d) from -50 to +50 cm⁻¹. (f) Raman spectra from 150 - 250 cm⁻¹ showing characteristic PtSe₂ modes. These spectra have been normalised to the E_g mode at ~ 180 cm⁻¹ for clarity.

layer-breathing mode (LBM), in-plane shear mode and standing wave mode. Fig. 5.10(e) shows low-frequency Stokes and anti-Stokes Raman spectra of the $PtSe_2$ flakes. This characteristic peak corresponds to LBM interlayer vibration in $PtSe_2$, which describes the motion of the top half and bottom half of the layers vibrating collectively but in opposite phase.^{144,146} The Stokes peaks are redshifted at ~ 22 cm⁻¹ from few-layer to thicker layers as the peaks approach 0 Raman shift, showing a clear trend in LBM peak position with increasing layer number.

To aid alignment for device fabrication, the flakes were exfoliated onto pre-patterned SiO₂/Si substrates with coordinates, allowing for ease of contacting *via* maskless UV photolithography (see Fig. 5.11(a) and (b)). Fig. 5.11(c) and (d) show the corresponding topography of typical flakes after contacting with Ti/Au. AFM determined the thickness of flake (c) to be 16 nm and (d) to be 13 nm which corresponds to ~ 27 and ~ 22 layers respectively.

Nanomachining was carried out on the contacted flake in Fig. 5.12(a), which consisted of \sim 15 layers



Figure 5.11: (a) and (b) are optical images of ME PtSe₂ flakes contacted *via* maskless UV photolithography. Contact pads are $100 \,\mu\text{m}^2$. (c) and (d) are corresponding topographical AFM images of flake devices, respectively (pre-machining).

(9 nm in thickness). Five machining passes with the same load of 2 μ N (250 mV) and tip velocity of 5 μ m/s were performed and the flake was imaged in between each pass. Fig. 5.12(b) shows the flake after the first nanomachining pass. The flake appears significantly cleaned of surface adsorbates, which are most likely from the ME process, rather than machining. The reduction in flake height was ~ 1.8 nm. Fig. 5.12(c) shows the device after the fifth machining pass. It is evident that significant destruction has occurred in the machined area with undesirable material removal particularly along the flake edges and grain boundaries. Fig. 5.12(d) shows a Raman map of the variation in position of the in-plane Eg (~ 173 cm⁻¹) mode of (c), as this can be influenced by layer thickness. From this, we can see the peak is at slightly greater wavenumbers in the nanomachined region, indicating the in-plane Eg mode. O'Brien *et al.* previously determined the thickness dependency of the PtSe₂ A_{1g} mode intensity (with respect to the intensity of the Eg mode),⁷⁶ similar to observations in other TMDs, such as MoS₂ and WS₂.^{151,269} We can see in Fig. 5.12(e) that the machined area, due to the lower ratio value, is indeed thinner than the pristine region, in agreement with (d). However, if we examine the variation in width of the E_g peak in a map (Fig. 5.12(f)), the E_g peak signal is at its broadest in the nanomachined area with a FWHM of $\sim 8 - 11 \text{ cm}^{-1}$ compared to $\sim 6 \text{ cm}^{-1}$ in the pristine area. Szydlowska *et al.* deemed E_g FWHM > 7 cm⁻¹ to be of lesser quality, indicating though thinner, the PtSe₂ layers are more disordered/less crystalline from the machining process.²³⁵ This would also suggest that successive passes of nanomachining on an area results in more defective/damaged material that is easier to nanomachine, leading to more material removed in those areas.



Figure 5.12: Topographical AFM images of ME PtSe₂ flakes (a) before nanomachining, (b) after first nanomachining pass and (c) fifth pass (white dashed line indicates machined area). Each pass had 2 μ N load and 20 μ m/s tip velocity. (d) Raman peak position map of (c) showing variation in PtSe₂ E_g position (173 cm⁻¹) after nanomachining. (e) Map of Raman mode intensity ratio A_{1g}/E_g. (f) Map of Raman E_g mode width. (g) Graph showing IV curve of the 2-probe measurement performed on PtSe₂ before and after nanomachining.

A two-probe method was used to electrically characterise the device in ambient conditions. A voltage sweep between -1 to 1 V was performed before and after nanomachining. The resulting I-V curve in Fig. 5.12(e) reveals an increase in conductivity post-nanomachining. This is in contrast to a potentially expected decrease in current indicating semiconductor behaviour (if <3L) or destruction of the device due to nanomachining. This discrepancy in conductivity can be most likely attributed

to varied conditions between measurements, such as changes in contact resistance due to the removal of adsorbates/contaminants.

Topographical information was extracted to gain a better understanding of the nanomachining process on PtSe₂ when performing the five passes in Fig. 5.12. Fig. 5.13(a) shows where on the AFM height retrace images the line profiles were extracted from. Fig. 5.13(b) are the topographical profiles parallel (left plot) and normal (right plot) to the machining direction. It is evident more material is removed incrementally with the increasing number of machining passes. However, the machining at constant load (2 μ N) appeared to reach a threshold at approx. the 8 nm mark (~ 6 nm in channel height relative to the SiO₂), whereby it cannot remove further material incrementally but rather tears/strips away layers in bulk. This led to complete removal of PtSe₂ in some areas during the 5th pass, which can be seen in Fig. 5.13(b). This is most likely due to PtSe₂'s strong interlayer interaction being greater than its adherence to SiO₂.



Figure 5.13: (a) Topographical AFM image of ME PtSe₂ flake after first nanomachining pass (2 μ N load, 20 μ m/s tip velocity). White dashed line indicates machined area with white arrows showing topographical line profiles and directions. (b) Topographical line profiles parallel and perpendicular to the machining direction.

The flake in Fig. 5.11(b)/(d) was subsequently contacted to the AFM puck to ensure a good conduction path (Fig. 5.14(a)) for conductive scanning probe measurements, i.e. KPFM and PF-TUNA. This was achieved by contacting one of the TiAu contact pads to the puck using Ag paint and an Ag wire (Fig. 5.14(b)). The flake was then nanomachined with a load of 0.36μ N.

However, this load was too small and appeared to only partially clean the flake surface of adsorbates/contaminants from the ME process (Fig. 5.14(c)). Fig. 5.14(d) is the corresponding KPFM image of the local surface potential/contact potential difference (CPD). It shows an increase in surface potential of 100 ± 60 mV in the machined area relative to the pristine PtSe₂ (+0.1 eV in φ_{PtSe_2}). This increase can be attributed to the removal of adsorbates which tend to lower the work function of the material surface.²⁰² PF-TUNA was performed on the same area at a sample bias of 1 V. Fig. 5.14(e) shows the topography, while (f) and (g) show the tunneling and peak currents respectively (see Chapter 2 Section 2.1.6 for breakdown of measured currents). We can see the 'cleaned' area appears significantly more conductive than the rest of the flake. However, again this is likely due to removal of adsorbates rather than improved conductivity from nanomachining as greater contrast is observed when the layer beneath can be seen rather than the defective remaining layers.



Figure 5.14: (a) Optical image of macro Ag wire contact on PtSe₂ flake device. (b) Zoom of macro contact. (c) Topography and (d) corresponding contact potential difference (CPD) of contacted flake (a) post-machining (1st pass, $0.36 \,\mu$ N load, $20 \,\mu$ m/s tip velocity). White dashed line indicates nanomachining area. (e) Topography, (f) tunneling current and (g) peak current maps of (a). (h) Graph showing IV curve of the 2-probe PtSe₂ before and after nanomachining.

Similar to the previous flake, a two-probe method was used to electrically characterise the device in ambient conditions. A voltage sweep between -2 to 2 V was performed before and after

nanomachining. The resulting I-V curve in Fig. 5.14(h) reveals a slight decrease in conductivity post-nanomachining but ultimately no significant change is evident as the flake is still exhibiting semi-metallic behaviour. However, in comparison to the previous PtSe₂ device (Fig. 5.12), the conductivity is much greater overall, probably due to Fig. 5.14 being an intact crystal, rather than segmented, with would decrease the conductivity. Again, these discrepancies in conductivity can also be attributed to varied conditions between measurements and devices, e.g. changes in contact resistance.

The flake was nanomachined further to remove layers. Multiple passes were repeatedly performed with increasing loads from 0.8 μ N (100 mV deflection setpoint) incrementally to 2.7 μ N (350 mV). In Fig. 5.15(a), we can see that the increased load of 0.8 μ N from the initial pass of 0.36 μ N allows for consistent cleaning of the flake surface and removal of adsorbates, resulting in a 2 nm reduction in the flake height. From the cleaned step height in Fig. 5.15(a), the flake is determined to be ~ 10.8 nm thick, corresponding to ~ 18 layers (L). Fig. 5.15(b) shows the same area after nanomachining with a 1.6 μ N load (200 mV deflection setpoint).

An average of the recessed height compared to the original flake height shows a reduction of ~ 5 nm in flake thickness. The recess average step height on SiO₂ is ~ 6.6nm (~ 11 L), at thickness at which the material should be semimetallic. Six more machining passes (nine in total) were performed with increasing setpoints; 200 mV (×3), 250 mV, 300 mV and 350 mV, which corresponded to 1.6, 1.95, 2.3 and 2.7 μ N, respectively, the last of which resulted in destruction of the device (Fig. 5.15(c) and (d)). Raman maps depicting the variation in FWHM of the in-plane E_g mode (177 cm⁻¹, Fig. 5.15(f)) and the ratio of the out-of-plane mode A_{1g} mode intensity (207 cm⁻¹) to the E_g intensity (Fig. 5.15(g)) of the PtSe₂ show the machined and destroyed regions, as well as the pile-up at the sides. The lack of signal in the destroyed region indicates that there is essentially no longer PtSe₂ present as it was ripped away during nanomachining. In addition, the FWHM broadening of the machined region would indicate some disordering of the PtSe₂ is occurring during nanomachining. However, the FWHM



Figure 5.15: Topographical AFM images of ME PtSe₂ flake (a) after 0.8μ N load, (b) after 1.6μ N load and (c) after 2.7μ N load (tip velocity constant for each pass at 20μ m/s). (d) Close-up of machined region in (c). (e) Optical image of (c). Red box indicates Raman scan area. (f) Raman peak FWHM map of (c) showing the variation in width of in-plane E_g mode (177 cm⁻¹). (g) Raman map of (c) showing the ratio of out-of-plane A_{1g} mode intensity (207 cm⁻¹) to E_g mode intensity (177 cm⁻¹). Dashed lines in (f) outline the Au contacts. (h) Raman spectra comparing E_g and A_{1g} modes. These spectra have been normalised to the E_g mode at ~177 cm⁻¹ for clarity.

remains $< 8.5 \text{ cm}^{-1}$, indicating less destruction of the PtSe₂ than in the previous case. In the ratio map (Fig. 5.15(g)), we can see the machined area to be of lower ratio of A_{1g} to E_g, indicating thinning of the PtSe₂. Fig. 5.15(h)) shows a comparison of the characteristic Raman spectra of the pristine, nanomachined and destroyed regions of the flake. The slight decrease in the relative height of the A_{1g} mode to the E_g mode between the pristine and machined areas show that the region was marginally thinned. Like the previous device, nanomachining appeared to cause some damage to the PtSe₂ layers during thinning but less damage appeared to occur at repeated loads of $< 2 \,\mu$ N.

Again, topographical information was extracted to gain a better understanding of the nanomachining process on $PtSe_2$ while performing the nine passes in Fig. 5.15. Fig. 5.16(a) shows the loads used in each nanomachining operation (tip velocity remained constant at 20 μ m/s). Fig. 5.16(b) is a comparison of topographical profiles of the $PtSe_2$ device surface after most iterations of nanomachining. Pass 6 (P6) to pass 9 (P9) were performed consecutively, without periodic



Figure 5.16: (a) Graph of machining loading forces vs. corresponding order of passes (tip velocity constant at $20 \,\mu$ m/s). (b) Topographical line profiles perpendicular to machining direction. (c) Plot of loading forces vs. resulting recess heights and material removed per pass.

imaging. Using the line profiles in (b), the average recess step height and subsequently the average amount of material removed with each pass were extracted and plotted in Fig. 5.16(c). There is an general upward trend of increasing passes and load results in more material being removed. Between loads of $0.8 - 1.6 \mu N (100 - 200 \text{ mV} deflection setpoint)$, an average of 1 nm of PtSe₂ is removed with each pass, which is equivalent to $\sim 1.5 \text{ L}$. For the last pass at 2.7 μN (P9), we see that the most material is removed (average $\sim 6 \text{ nm}$), however inconsistencies in the layer removal become evident, with some areas further recessed than others, indicating cutting and stripping of the PtSe₂ layers.

5.4 CONCLUSION

A novel technique for exploring and manipulating TMD layers has been developed *via* nanomachining using AFM. This technique was used to reduce layer thickness of $PtSe_2$ TAC films and ME flakes. The machining is required to isolate the thinnest section from the bulk in the hopes of contacting it to observe mobilities, amongst other properties. AFM and SEM results revealed the nanomachining of TAC films to be quite difficult due to the roughness of the films post-selenisation, but can be used to produce thinner, smoother films overall. Machining of pristine

Pt channels allowed more control over the quantity of nanomachining, and also quality. However, selenisation of more than 5 nm Pt results in disruption of the films due to exponential increase in edge-rich PtSe₂, thus significantly decreasing electronic transport in plane and reducing the reliability of fabrication for potential devices. For comparison with TAC films, pristine ME PtSe₂ flakes were nanomachined with a stepped depth accuracy of ~ 0.8 nm, approximately the thickness of one ML, showing the capability of layer-by-layer removal via this technique. Further in-depth characterisation was conducted on flakes of varying thicknesses using AFM and low-frequency Raman spectroscopy. Using maskless UV lithography, two terminal devices were created from ME flakes. The devices were subsequently nanomachined and the results examined. Initially, it appeared that nanomachining cleaned the flake surface of adsorbates and contaminants, most likely from the mechanical exfoliation and ambient conditions. The cleaning occurred at nanomachining forces $<1.5 \,\mu$ N. With forces $>1.5 \,\mu$ N, the flakes were marginally thinned but the layers tended to adhere together rather than separate incrementally. This is most likely due to the extraordinarily strong interlayer coupling in PtSe₂. In addition, nanomachining at forces $>2.3 \,\mu$ N resulted in complete material removal from the substrate, due to the weak coupling between the ME PtSe₂ and SiO₂, particularly in comparison with the PtSe₂ interlayer interaction. Despite this, there is scope for nanomachining as a thinning method for devising a 'self-contacting' PtSe₂ FET, should a more appropriate substrate be selected, such as Au, where the 2D crystal/Au binding is invariably stronger the corresponding interlayer bonding.²⁵² However, an Au substrate would present challenges for device measurement. In addition, there is potential to apply these techniques to other TMD systems like PtS₂, which has very similar properties to PtSe₂, such as crystal structure, interlayer interactions and electronic structure.^{78,81,234,270}

Science makes people reach selflessly for truth and objectivity; it teaches people to accept reality, with wonder and admiration, not to mention the deep awe and joy that the natural order of things brings to the true scientist.

Lise Meitner, physicist

6

Conclusions & Future Work

For over 400 years, the optical microscope has reigned supreme in the field of microscopy. Nevertheless other types of microscopy techniques based on electrons and atomic forces have come to the fore in the last century. In particular, SPM has revolutionised nanotechnology and allowed the study and manipulation of materials on the nanoscale, making it ideal for the study of solid-state physics and semiconductor technologies. As the research field of 2D materials continues to expand exponentially, particularly with applications in future nanoelectronics, making the case for the need of an SPM-based toolbox, such as AFM, whose capability as both a microscopic technique and nano-manipulator for these materials has never been more relevent.

In this thesis, various tools and techniques for characterising and manipulating several TMD materials and heterostructures have been examined and developed. Particular focus was given to AFM and KPFM of VPD-grown and ME TMDs and AFM-based mechanical lithography of these materials, namely nanoshaving and nanomachining.

In Chapter 4, the non-covalent functionalisation of monolayer TMDs, MoS₂ and WS₂, on Si/SiO₂ with organic SAMs of PDI was explored. These organic-inorganic heterostructures were manipulated using the approach of nanoshaving and the reversibility of such functionalisation was demonstrated using the complementary non-destructive techniques of KPFM and Raman spectroscopy. Nanoshaving pristine monolayer TMD flakes had negligible effect on the topography but KPFM showed contrast in CPD between the shaved and non-shaved areas, most likely due to clearing of surface contaminants. It was shown that PDI selectively forms SAMs on TMD surfaces rather than Si/SiO₂. A grating pattern with feature sizes < 250 nm was successfully shaved into the PDI/MoS₂ heterostructure, demonstrating the high-resolution capabilities of nanoshaving as a lithographic technique. These results indicate that nanoshaving is a viable lithography technique that introduces minimal contaminants, i.e. no resist necessary and without collateral damage to the SAM. Nanoshaving could be suitable for the fabrication of other ordered optical/photonic structures, e.g. by rotating the sample 90°, a shaving pass of the same spacing parameters would produce a dot array. In addition, non-covalent co-functionalisation of MoS₂ was demonstrated PTCDA SAMs on previously nanoshaved areas, using allowing more complex patterns/arrangements to be produced.

Chapter 5 reported a technique for exploring and manipulating TMD layers *via* nanomachining using AFM. This technique was used to reduce layer thickness of TAC-grown PtSe₂ films and ME flakes. AFM and SEM results revealed the nanomachining of TAC-grown films to be quite difficult

due to the roughness of the films post-selenisation when compared to exfoliated materials. Nevertheless nanomachining TAC films resulted in thinner, smoother films overall. Machining of pristine Pt channels allowed more control over the quantity of nanomachining, and also quality. For comparison with TAC-grown films, pristine ME PtSe₂ flakes were nanomachined with further in-depth characterisation conducted on flakes of varying thickness using AFM and low-frequency Raman spectroscopy. Using maskless UV lithography, two terminal devices were created from ME flakes which were subsequently nanomachined. Nanomachining at forces $>3 \mu N$ resulted in complete material removal from the substrate due to the weak coupling between the ME PtSe₂ and SiO₂, particularly in comparison with the PtSe₂ interlayer interaction. For weaker forces, it was also found that successive passes of nanomachining on an area resulted in more defective/damaged material that was easier to nanomachine, leading to more material removed in those areas.

Despite this, there is scope for nanomachining as a thinning method for devising a 'self-contacting' PtSe₂ FET, should a more appropriate substrate be selected, such as Au, where the 2D crystal/Au binding is invariably stronger than the corresponding interlayer bonding.²⁵² However, an Au substrate would present challenges for device measurement. In addition, there is potential to apply these techniques to other TMD systems like PtS₂, which has very similar properties to PtSe₂, such as crystal structure, interlayer interactions and electronic structure.^{78,81,234,270} This suggests a path for continuous monitoring of device performance with each layer removal down to the monolayer, starting with metallic multi-layers and thinning down to a semiconducting monolayer. This would enable the design of 'self-contacted' devices based on TMDs through the creation of a semiconducting channel *via* nanomachining with high mobility, low contact resistance and low power.

For nanoshaving non-covalently functionalised SAMs on TMDs, there is scope for exploring many possibly combinations of SAMs or other adhesive molecules or even other TMDs as heterostructures. Within this, there is opportunity to delve into the fabrication and measurement of

p-n junctions, particularly using KPFM as a technique.^{271,272} Additonally, to better observe the PL components, further measurements should be carried out on the heterostructures at lower temperatures (e.g. 77 K) to reduce non-radiative recombination effects on the spectrum. With regards nanoshaving's capability of nanolithography, the deposition of materials on the nanoshaved patterns should be examined as well as more advanced structures such as dot arrays. These nanoscale dot arrays could provide an excellent platform towards quantum information processing.²⁷³

To conclude, the techniques of AFM-based mechanical manipulation, namely nanoshaving and nanomachining, of TMDs and TMD-based heterostructures have been explored and examined and remarkable control over material removal and patterning at the nanoscale has been shown. The properties of both pristine and manipulated TMDs and TMD heterostructures have been thoroughly studied by surface microscopy and spectroscopic techniques, demonstrating the capability of an AFM-based toolbox for examining 2D materials. Nevertheless, there is further scope for more in-depth study and characterisation of the materials as well as opportunities to further improve manipulation parameters.


Nanoshaving Raman Spectra



Figure A.1: Nanoshaved pristine MoS_2 Raman — (a) Optical image of the MoS_2 area scanned using Raman spectroscopy (red box) with shaved area indicated by white dashed box. (b) Raman peak intensity map of MoS_2 out-of-plane A_{1g} (406 cm⁻¹) mode. (c) average Raman spectra comparison of shaved and non-shaved pristine MoS_2 showing both E_{2g}^1 and A_{1g} modes (normalised to Si peak).



Figure A.2: Nanoshaved PDI/MoS₂ heterostructure Raman – (a) Optical image of the PDI/MoS₂ heterostructure area scanned using Raman spectroscopy (red box) with shaved area indicated by white dashed box. (b) Raman peak intensity map of MoS₂ out-of-plane A_{1g} (406 cm⁻¹) mode. (c) Raman peak intensity map of PDI (~1300 cm⁻¹) mode. (c) average Raman spectra comparison of shaved and non-shaved PDI/MoS₂ showing both E_{2g}^1 and A_{1g} modes (normalised to Si peak).



Figure A.3: PDI Photoluminescence/Raman — (a) Raman peak intensity map of PDI (\sim 1585 cm⁻¹) mode. (b) PL peak intensity map of PDI (\sim 2.12 eV). (c) average PL spectra comparison of PDI on SiO₂, shaved MoS₂ and non-shaved PDI on MoS₂.



Figure A.4: Nanoshaved PDI/WS₂ Heterostructure Raman – (a) Optical image of the PDI/WS₂ area scanned using Raman spectroscopy (red box) with shaved area indicated by white dashed box. (b) average Raman spectra comparison of shaved and non-shaved PDI/WS₂ 2LA(M) + E_{2g}^1 mode (352 cm⁻¹, normalised to Si peak). (c) average Raman spectra comparison of shaved WS₂ and non-shaved PDI on WS₂ showing characteristic PDI peaks.

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